

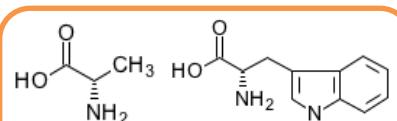
Bases de données en métabolomique

Christophe Junot

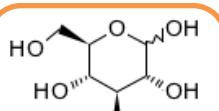
CEA/Laboratoire d'Etude du Métabolisme des Médicaments
CEA-Saclay (iBiTec-S)
christophe.junot@cea.fr

Metabolites and metabolome

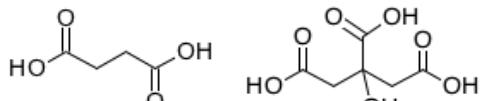
Primary metabolites



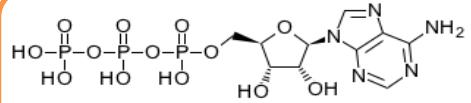
Aminoacids



Sugars



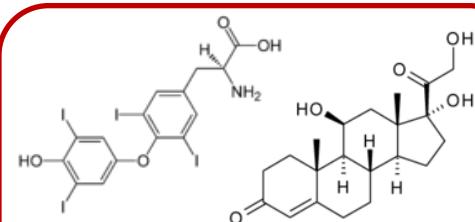
Organic acids



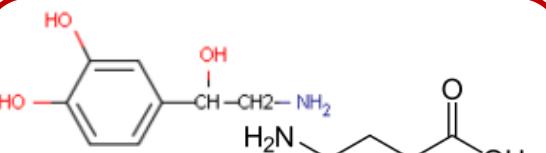
Nucléotides

(...)

Secondary metabolites



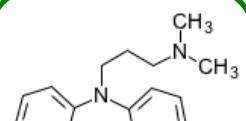
Hormones



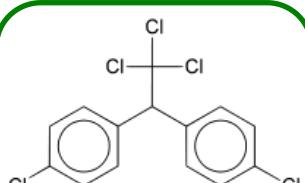
Neurotransmitters

(...)

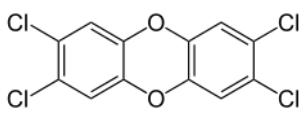
Xenobiotics



Drugs



Pesticides



Pollutants

(...)

Food / drinking

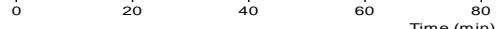
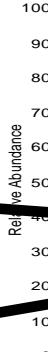
Metabolic fingerprint

Central metabolism

Gut microbiota

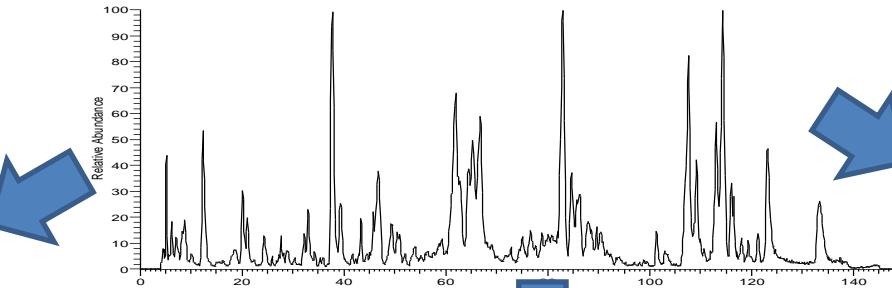
Environment :
xenobiotics
(pollutants, drugs...)

Pathology



How to detect metabolites in biological media?

Metabolic fingerprint



NMR

- Simple, non invasive
- Rapid
- Robust: analysis of large series of samples
- But:
- Limited sensitivity

GC-EI-MS

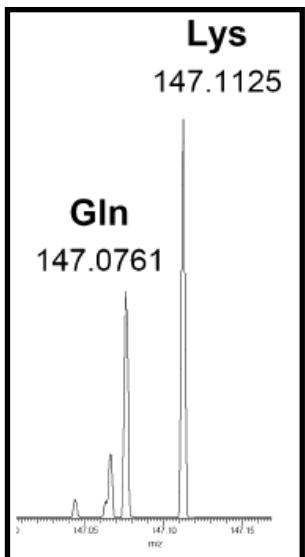
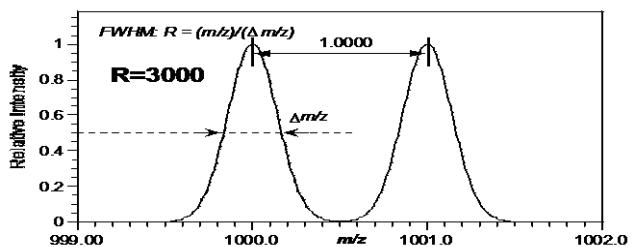
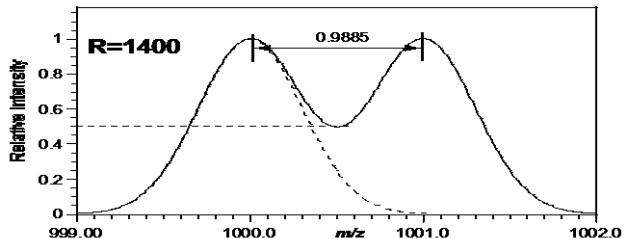
- Sensitive
- Reproducible
- Spectral libraries
- But:
- Chemical derivatization of non volatile compounds
- Issue of thermolabile compounds

LC-API-MS

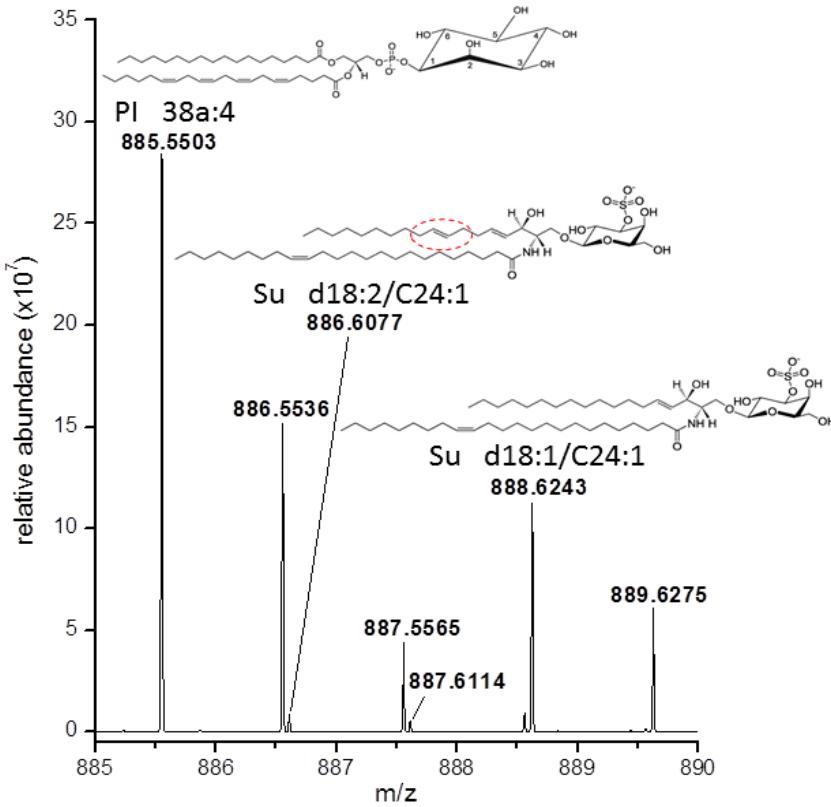
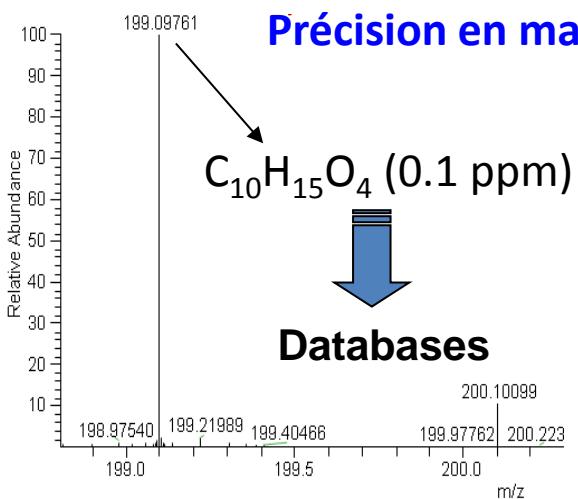
- Molecular mass of intact compounds
- Analysis of thermolabile compounds
- sensitive
- But:
- Poor inter-platform reproducibility

MS à haute résolution: détecter plus de métabolites et les identifier plus facilement

Résolution en masse

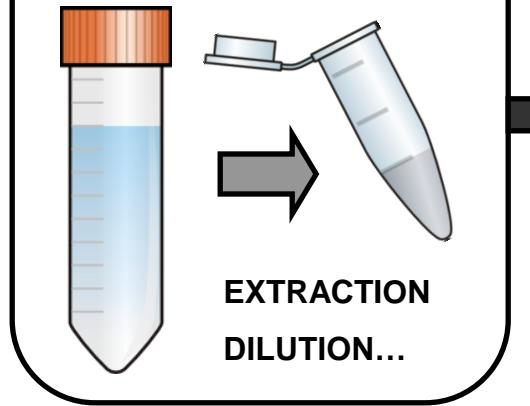


Précision en masse

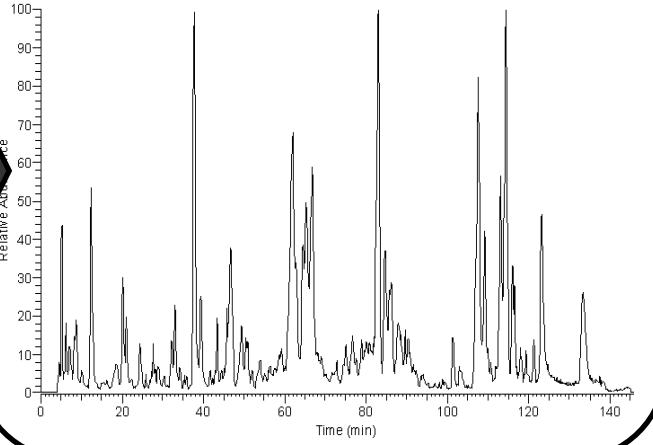


Déroulement d'une analyse métabolomique

PRÉPARATION DES ÉCHANTILLONS



ACQUISITION DES EMPREINTES



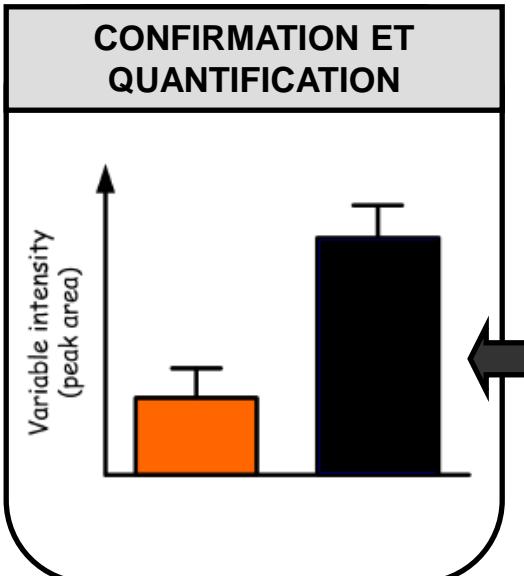
TRAITEMENT DES DONNÉES

DÉTECTION AUTOMATIQUE DES SIGNAUX

ECH.

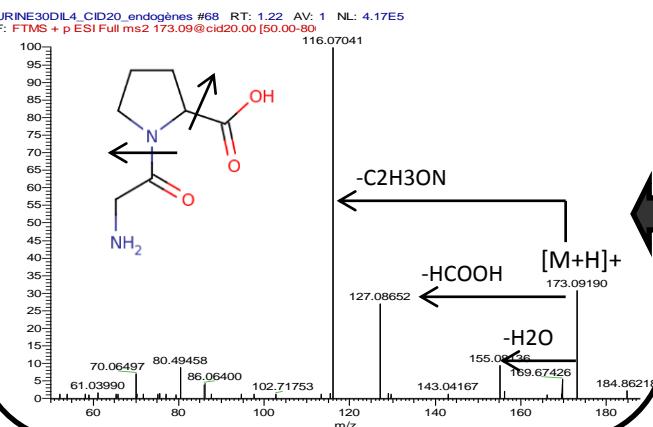
Peak No	Retention	Mass	J5-T						
1	13.24	100	1.8E+03	1.5E+03	2.4E+03	1.7E+03	2.6E+03	1.0E+03	7.0E+02
2	13.98	100	1.8E+03	1.5E+03	2.0E+03	1.7E+03	2.0E+03	1.3E+03	1.5E+03
3	42.25	106	2.0E+03	4.3E+03	2.9E+03	3.5E+03	1.8E+03	2.0E+03	2.0E+03
4	46.01	131	1.8E+03	2.0E+03	1.8E+03	1.8E+03	1.8E+03	1.8E+03	1.8E+03
5	16.92	114	2.0E+03	3.7E+04	4.5E+04	2.1E+05	8.3E+04	2.5E+04	2.5E+04
6	17.26	114	2.0E+03	3.7E+04	4.5E+04	1.6E+05	7.2E+04	2.5E+04	2.5E+04
7	17.41	124	2.0E+03	3.7E+04	4.5E+04	1.4E+05	6.4E+04	2.5E+04	2.5E+04
8	18.01	114	2.0E+03	3.7E+04	2.9E+04	1.0E+04	4.3E+04	2.5E+04	2.5E+04
9	4.19	126	9.4E+03	1.6E+03	6.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
10	4.46	126	2.1E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
11	4.93	126	1.8E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
12	5.07	126	1.8E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
13	5.40	126	1.3E+03	1.6E+03	1.1E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
14	5.85	126	1.3E+03	1.6E+03	1.1E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
15	6.32	126	1.5E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
16	18.56	126	1.9E+03	1.6E+03	2.0E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
17	11.05	126	1.9E+03	1.6E+03	2.0E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
18	11.33	126	1.6E+03	1.6E+03	1.5E+03	9.1E+03	2.3E+03	1.2E+04	1.2E+04
19	11.80	126	2.1E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
20	16.36	126	1.6E+03	1.6E+03	2.1E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
21	9.13	126	1.9E+03	1.6E+03	2.0E+03	1.7E+04	2.3E+04	1.2E+04	1.2E+04
22	4.39	143	2.9E+03	2.4E+03	2.8E+03	2.9E+03	2.9E+03	2.5E+03	3.2E+03
23	5.07	143	2.3E+03	2.4E+03	2.5E+03	2.3E+03	2.3E+03	1.9E+03	2.3E+03
24	5.81	143	2.3E+03	2.4E+03	2.5E+03	2.3E+03	2.3E+03	1.9E+03	2.3E+03
25	26.39	143	1.3E+03	8.3E+02	5.8E+03	2.3E+03	2.3E+03	1.2E+03	1.2E+03
26	6.58	153	4.0E+03	1.1E+03	8.9E+03	9.9E+03	3.4E+03	4.5E+03	4.5E+03
27	7.01	153	4.0E+03	1.1E+03	8.9E+03	9.9E+03	3.4E+03	4.5E+03	4.5E+03
28	6.72	154	4.2E+03	5.0E+03	2.9E+03	2.0E+04	3.1E+03	3.0E+03	3.0E+03
29	6.98	154	5.3E+03	2.0E+03	6.3E+03	2.0E+04	2.8E+03	2.1E+03	2.1E+03
30	7.24	154	2.3E+03	6.7E+03	2.3E+03	1.0E+03	2.8E+03	2.3E+03	2.3E+03
31	17.54	159	2.3E+03	6.7E+03	2.3E+03	3.0E+03	2.7E+03	2.3E+03	2.3E+03
32	17.87	159	2.3E+03	6.7E+03	2.3E+03	2.8E+03	2.7E+03	2.3E+03	1.1E+03
33	18.42	159	2.3E+03	6.7E+03	2.3E+03	2.8E+03	2.7E+03	2.3E+03	3.0E+03

CONFIRMATION ET QUANTIFICATION

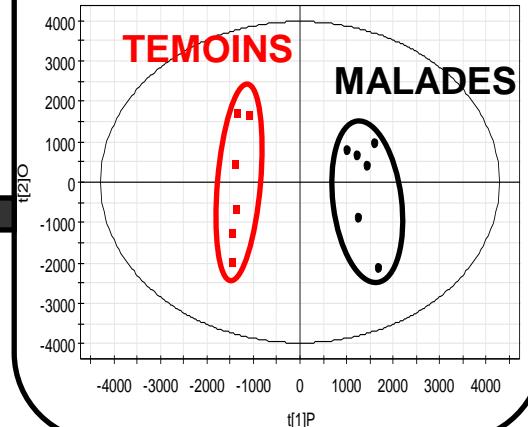


IDENTIFICATION DES BIOMARQUEURS

BASES DES DONNÉES, MS/MS ...



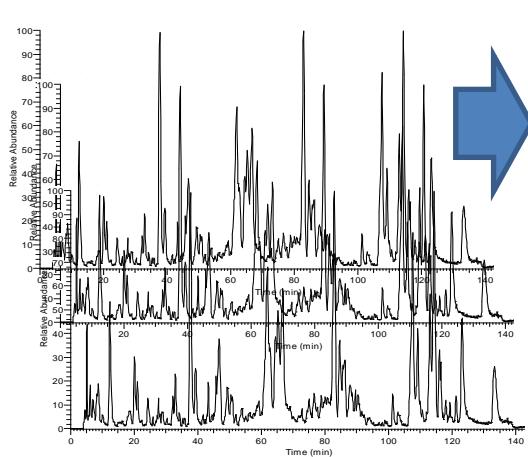
DÉTECTION AUTOMATIQUE DES SIGNAUX



Les différents types de bases de données en métabolomiques

Bases de données biochimiques/métaboliques

annotation



Variables (Rt-mass)

Peak Nr	Ret(min)	Mass	J5-T	J5-T	J5-T	J5-T	J5-T	J5-T
1	13.24	100	1.8E+03	1.5E+03	2.4E+05	1.7E+03	2.0E+03	7.9E+04
2	13.38	100	1.8E+03	1.5E+03	2.0E+03	1.7E+03	2.0E+03	1.5E+04
3	42.25	106	2.0E+05	4.3E+04	2.9E+05	3.5E+04	1.8E+05	2.0E+03
4	16.65	114	2.0E+03	3.7E+04	4.5E+04	1.0E+04	3.9E+05	2.5E+04
5	16.92	114	2.0E+03	3.7E+04	4.5E+04	2.1E+05	8.3E+04	2.5E+04
6	17.26	114	2.0E+03	3.7E+04	4.5E+04	1.6E+05	2.5E+05	7.2E+05
7	17.54	114	2.0E+03	3.7E+04	2.3E+05	1.4E+05	4.3E+04	2.5E+04
8	18.01	114	2.0E+03	3.7E+04	2.8E+05	1.0E+04	4.3E+04	2.5E+04
9	4.19	126	9.4E+04	1.5E+03	6.4E+04	1.7E+04	2.3E+03	1.2E+04
10	4.66	126	1.4E+05	1.6E+03	1.3E+05	1.7E+04	2.3E+03	1.2E+04
11	4.93	126	2.1E+03	1.6E+03	2.4E+03	1.7E+04	2.3E+03	1.2E+04
12	5.07	126	1.8E+05	2.0E+03	2.4E+03	1.7E+04	2.3E+03	1.5E+05
13	5.40	126	1.3E+05	1.6E+03	1.1E+05	1.7E+04	2.3E+03	2.2E+04
14	5.86	126	2.1E+03	1.6E+03	2.0E+05	9.1E+03	2.3E+03	4.9E+04
15	6.32	126	1.5E+04	1.6E+03	2.4E+03	1.7E+04	2.3E+03	1.2E+04
16	10.56	126	1.9E+05	1.6E+03	2.0E+05	9.1E+03	2.3E+03	1.2E+04
17	11.05	126	2.1E+03	1.6E+03	2.4E+03	9.1E+03	2.3E+03	1.2E+04
18	11.33	126	1.0E+05	1.6E+03	1.5E+05	9.1E+03	2.3E+03	1.2E+04
19	11.80	126	2.1E+03	1.6E+03	2.4E+03	9.1E+03	2.3E+03	1.2E+04
20	16.36	126	2.0E+03	1.6E+03	2.4E+04	9.1E+03	2.3E+03	1.2E+04
21	9.04	138	1.9E+03	3.9E+04	2.1E+04	1.4E+04	2.3E+03	2.7E+04
22	4.39	143	2.9E+05	2.4E+05	2.8E+03	2.9E+05	2.9E+04	2.5E+05
23	5.07	143	2.3E+03	2.1E+03	2.5E+05	2.3E+03	1.9E+05	3.2E+05
24	5.20	143	5.5E+05	2.1E+03	1.1E+05	3.6E+04	2.9E+04	3.4E+05
25	26.39	143	1.3E+04	8.3E+04	5.8E+04	2.3E+03	2.9E+04	1.2E+05
26	6.58	153	4.0E+04	1.1E+06	8.9E+04	9.9E+04	3.4E+04	4.5E+04
27	7.12	153	4.0E+04	2.6E+03	8.9E+04	6.1E+04	1.9E+05	3.7E+04
28	6.72	154	4.2E+05	5.0E+05	2.9E+03	2.0E+04	3.1E+03	3.9E+04
29	6.98	154	5.3E+05	2.0E+03	6.3E+05	2.0E+04	2.8E+05	2.1E+05
30	7.66	154	2.2E+03	6.7E+04	2.9E+03	2.0E+04	2.8E+05	2.1E+04
31	17.54	159	2.3E+03	2.0E+03	2.8E+03	3.0E+05	2.7E+03	3.8E+04
32	17.87	159	2.3E+03	2.0E+03	2.8E+03	2.8E+04	2.7E+03	1.1E+05
33	18.42	159	2.3E+03	2.0E+03	2.8E+03	2.8E+03	4.2E+05	2.7E+03

Bases de données spectrales:

ESI/MS: annotation

Samples

?

?

?

?

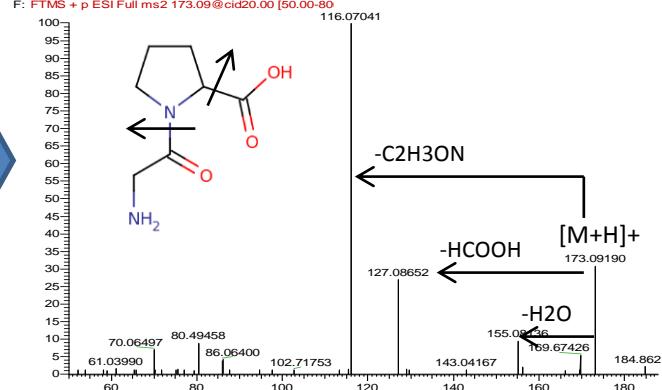
?

?

?

?

URINE30DIL4_CID20_endogènes #68 RT: 1.22 AV: 1 NL: 4.17E5



Patients with unexplained encephalopathy	
sugar acid	N-acetyluramic acid
Carbohydrates	Deoxyribose
	Pentose
	Mannitol or isomers
Hydroxy acids	Threonic acid
	Quinic acid
nucleoside derivative	Succinyladenosine
	Proribotide
Aminoacid and derivatives	Glutamic acid
	Am inoplastic acid
	N-acetyl-L-glutamic acid
	N-Acetyl-D-alanine
Pyrimidine derivated	Dihydroorotic acid
	Acetyl-carnitine
	Propionyl-carnitine
	Butyryl-carnitine
	Methylbutyryl-carnitine
Bile acids	Glycochenodeoxycholic acid
	Glycocholic acid

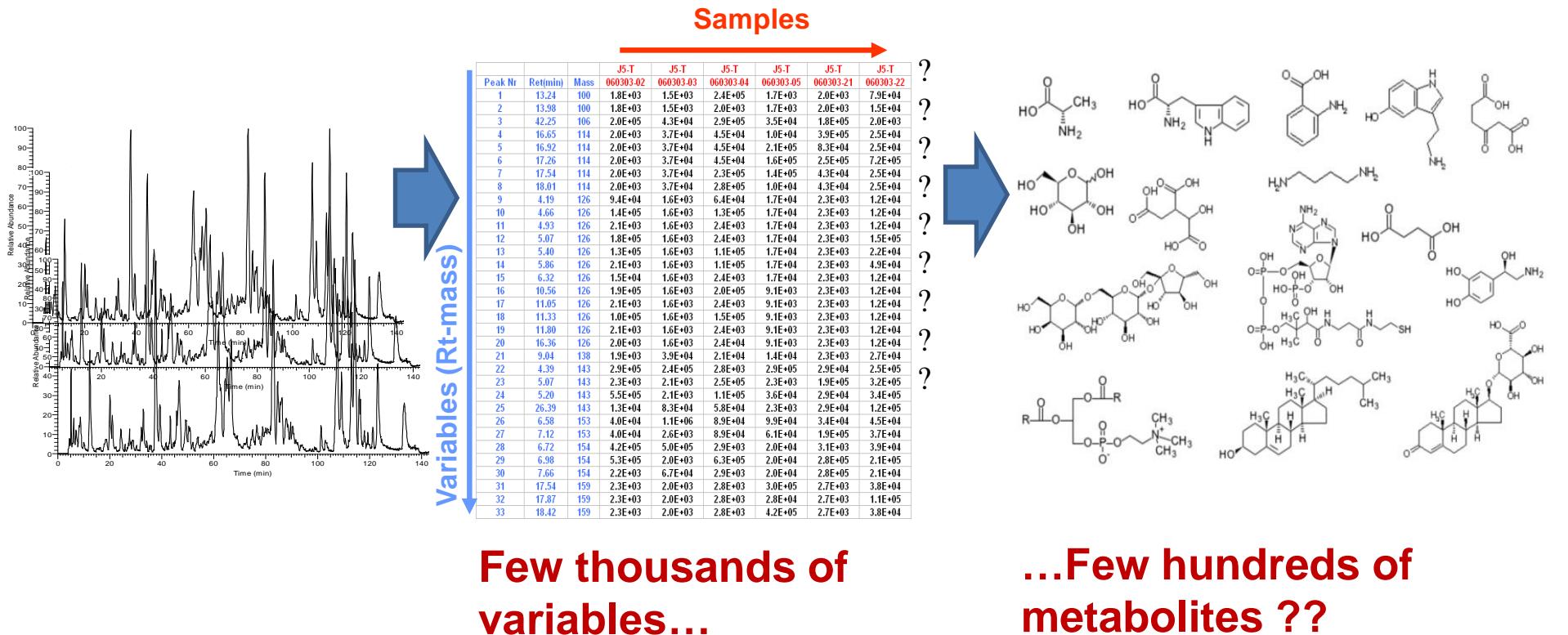
Bases de données de profils métaboliques

Spectral databases

database		thematic	Conception / URL	Instrument
NIST	◎	general	National institute for standard and technology (USA) www.nist.gov/srd/nist1a.htm	GC/MS
Fiehn Library	●	general	Fiehn Laboratory Univ California Davis – Genome center http://fiehnlab.ucdavis.edu/Metabolite-Library-2007	GC/MS
Golm	●	plant	Max Planck Institute for Molecular Plant Physiology (Germany) csbdb.mpimp-golm.mpg.de	GC/MS
HMDB	●	human metabolites	Department of Computing Science, University of Alberta (Canada) www.hmdb.ca/extrIndex.htm	NMR, API/MS/MS
Lipidmaps	●	lipidomics	LIPID MAPS Bioinformatics Core (USA) www.lipidmaps.org/data/index.html	API-MS/MS
Massbank	●	general	Keio university, university of Tokyo, Kyoto university, RIKEN plant Science center (Japan) and others www.massbank.jp	API-MS/MS
Metlin	●	human metabolites	Scripps Center for Mass Spectrometry metlin.scripps.edu	API-MS/MS
Brucker	○	general		NMRS
Madison Metabolomic Consortium database	●	general	http://mmcd.nmr.fam.wisc.edu/	NMRS

● free access, ◎ partially free access, ○ licenced

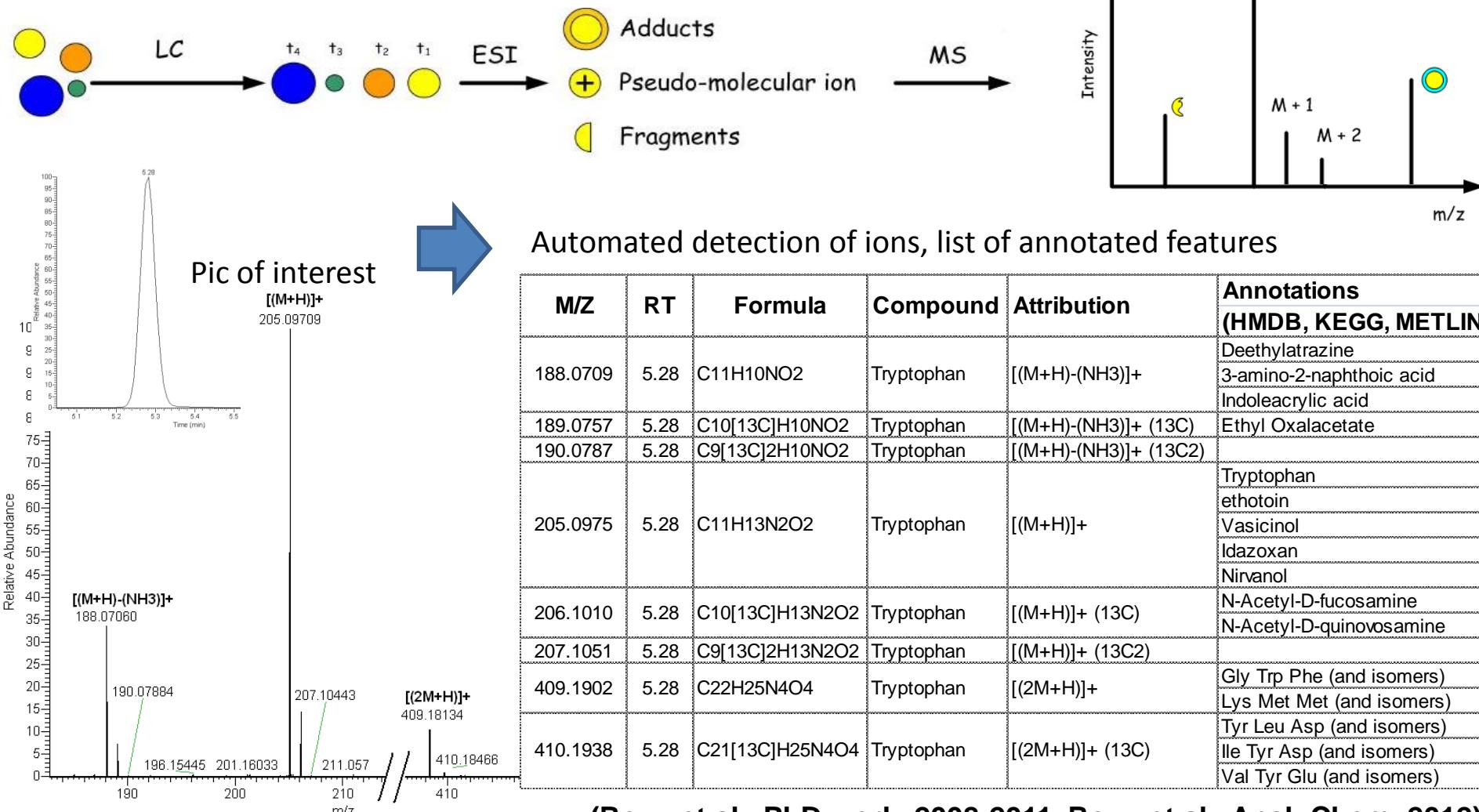
Annotation of peak lists is required to help for metabolite identification



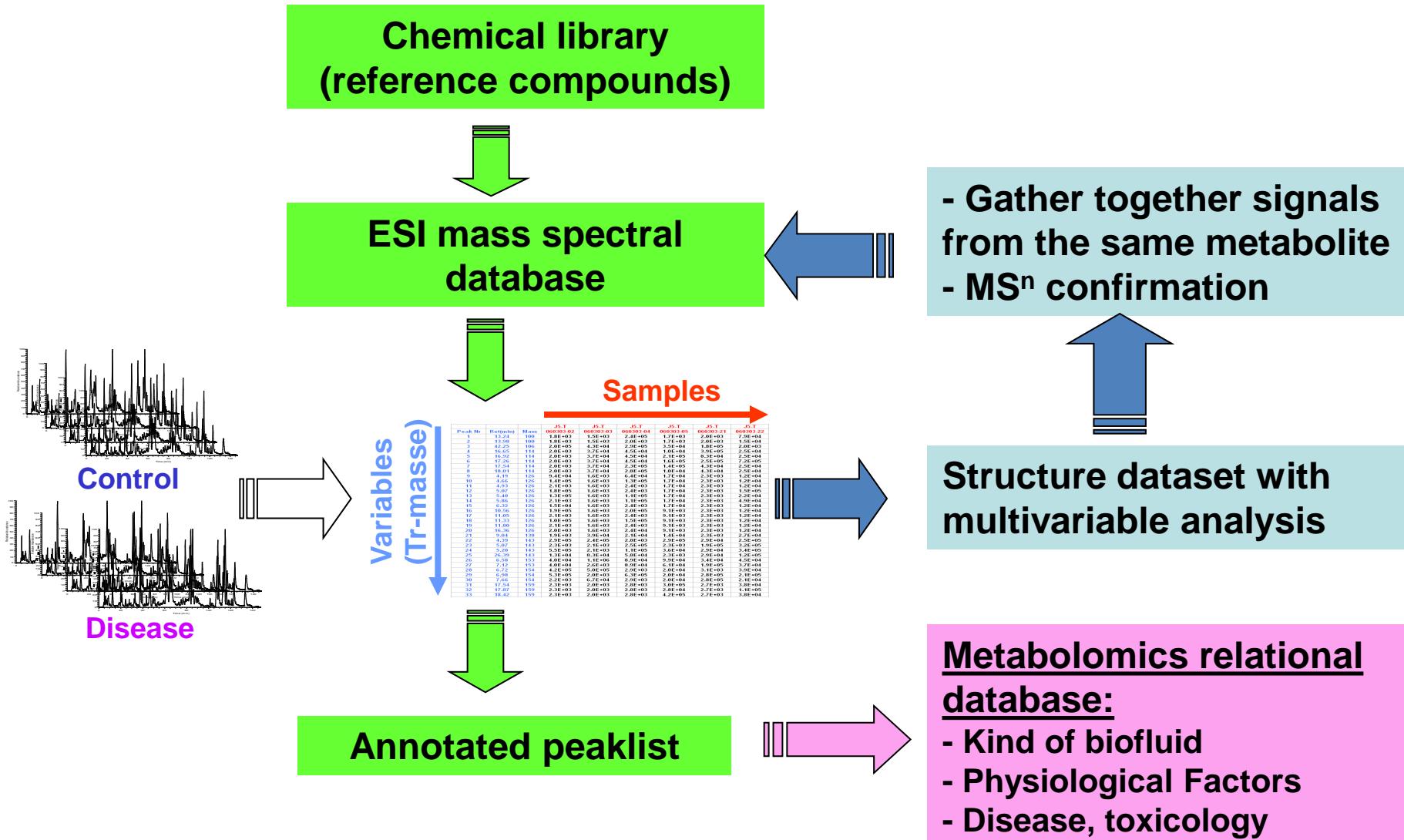
- Chemical and biochemical databases: KEGG (www.genome.jp/kegg), Metlin (www.metlin.scripps.edu), HMDB (www.hmdb.ca)
- spectral databases

The relevance of a spectral database

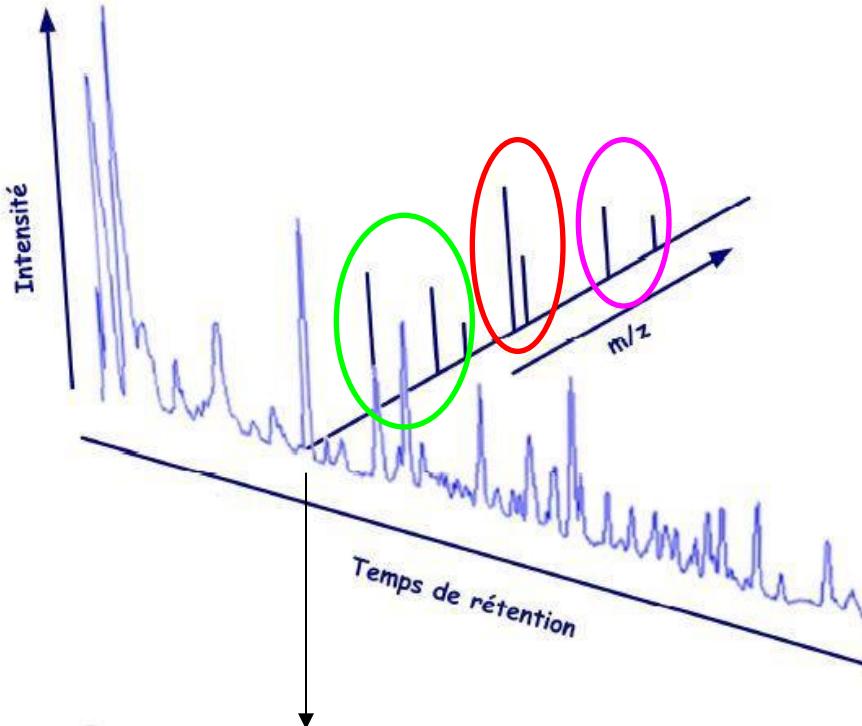
One molecule = several ions



Development of an ESI-mass spectral database for metabolomics : methodology



Analysis of reference compounds



One molecule = one retention time obtained by LC/MS

One molecule = many ions :
- Pseudo-molecular and isotopes
- Adducts
- Fragments

Annot. FIA*

M/Z	Intensity	Relative	M/Z (theo)	Delta (ppm)	RDBE	Composition	Attribution	FIA	UPLC (C8)	UPLC (C18)	HSFS
72.04431	7.7E+04	0.82	72.04439	-1.11	1.5	C3 H6 O N	[M+H]+(C6H10O3)-(H2O)+	0	1.80	4.77	11.17
90.05482	3.6E+05	3.78	90.05496	-1.52	0.5	C4 H8 O N	[M+H]+(C6H12O3)+(H2O)+	0	1.80	4.77	11.17
118.07424	2.1E+04	0.01	118.07432	-2.22	-2.0	C5 H10 O N	[M+H]+(C6H14O3)+(H2O)+	0	1.80	4.77	11.17
142.08585	1.0E+03	0.01	142.08626	-2.85	-2.5	C7 H12 O 2 N	[M+H]+(C8H16O3)+(H2O)+	0	1.80	4.77	11.17
174.11217	1.6E+04	0.17	174.11247	-0.3	1.5	C8 H16 O 2 N	[M+H]+(C9H20O3)+(H2O)+	0	1.80	4.77	11.17
184.09887	2.4E+03	0.03	184.09982	0.26	3.5	C9 H14 O 3 N	[M+H]+(HCOOH)+	0	1.80	4.77	11.17
202.10729	1.5E+04	0.16	202.10738	-0.46	2.5	C9 H16 O 4 N	[M+H]+(H2O)+	0	1.80	4.77	11.17
220.11799	9.6E+06	100	220.11795	0.16	1.5	C9 H16 N 0 5	[M+H]+	0	1.80	4.77	11.17
221.12135	6.8E+05	7.06	221.121305	0.2	1.5	C8 I3C H18 N 0 5	[M+H]+(13C)	0	1.80	4.77	11.17
222.12166	4.5E+04	0.47	222.122194	-2.4	1.5	C9 H18 N 0 4 18O	[M+H]+(18O)	0	1.80	4.77	11.17
223.12604	2.2E+03	0.02	223.125549	2.2	1.5	C8 I3C H18 N 0 4 18O	[M+H]+(13C+18O)	0	1.80	4.77	11.17
238.12923	1.7E+03	0.02	238.128515	3	0.5	C9 H20 N 0 4	[M+H]+(H2O)+	0	1.80	4.77	11.17
242.09585	6.8E+05	6.8E+05	242.09605	0.4	1.5	C9 H16 N Na 0 5	[M+Na]+	0	1.80	4.77	11.17
243.10331	4.4E+05	4.58	244.10325	0.25	1.5	C9 H17 N Na 0 5	[M+Na]+(13C)	0	1.80	4.77	11.17
244.10363	1.7E+04	0.01	244.104139	-2.09	1.5	C9 H17 N Na 0 4 18O	[M+Na]+(18O)	0	1.80	4.77	11.17
258.07386	1.2E+05	1.33	258.073833	0.1	1.5	C9 H17 K N 0 5	[M+K]+	0	1.80	4.77	11.17
259.07731	1.4E+03	0.02	259.077188	0.47	1.5	C8 I3C H17 K N 0 5	[M+K]+(13C)	0	1.80	4.77	11.17
264.08082	1.6E+05	1.74	264.08184	0.61	1.5	C9 H16 N Na 0 5	[M+Na]+(2Na)+	0	1.80	4.77	11.17
265.08498	3.0E+03	0.01	265.085195	-0.81	1.5	C8 I3C H16 N Na 0 5	[M+H+2Na]+(13C)	0	1.80	4.77	11.17
310.08654	1.1E+03	0.01	310.08732	-2.52	1.5	C10 H18 N 0 2 0 7	[M+Na]+(HCOONa)+	0	1.80	4.77	11.17
461.21113	6.4E+05	6.65	461.210569	1.22	2.5	C10 H18 N 0 2 0 10	[2M+Na]+	0	1.80	4.77	11.17
462.21438	1.0E+05	1.11	462.213924	0.99	2.5	C17 I3C H34 N 0 2 0 10	[2M+Na]+(13C)	0	1.80	4.77	11.17
463.2174	9.7E+03	0.1	463.217279	0.26	2.5	C16 I3C H34 N 0 2 0 10	[2M+Na]+(13C2)	0	1.80	4.77	11.17
483.19256	4.2E+04	0.43	483.192514	0.1	2.5	C18 H33 N 0 2 0 10	[2M+H+2Na]+	0	1.80	4.77	11.17

ESI-mass spectral database
→ Annot. SPI*

Reference compounds analysis :

FIA spectrum annotation

SPI-LIMS

Annotations	Annotation FIA
1. FIA 2. PDBA 3. SPI	Molécule: <input type="text" value="C7 H19 N3"/> Polarité: <input checked="" type="radio"/> Positive <input type="radio"/> Négative Décalage (ppm): <input type="text" value="0"/> Précision (\pm ppm): <input type="text" value="5"/> Intensité relative ZERO: <input checked="" type="radio"/> OUI <input type="radio"/> NON Fichier MS: <input type="text"/> <input type="button" value="Parcourir..."/> <input type="button" value="Annoter"/>
Peaklists Outils	

« Annotation FIA »: to calculate precise m/z of potential ions for a given mass formula and compare them to experimental m/z :

- Isotopes: ^{13}C , ^{34}S , ^{18}O ...
- Adducts: Na, K, Cl...
- Fragments (in source CID)

Reference compounds analysis

Pantothenic acid $[M+H]^+$

M/Z	Intensity	Relative	M/Z (theo)	Delta (ppm)	RDBE	Composition	Attribution	Retention time	Retention time	Retention time
72.04431	7.7E+04	0.82	72.04439	-1.11	1.5	C3 H6 O N	$[(M+H)-(C_6H_{10}O_3)-(H_2O)]^+$	1.80	4.77	11.17
90.05482	3.6E+05	3.78	90.05496	-1.55	0.5	C3 H8 O2 N	$[(M+H)-(C_6H_{10}O_3)]^+$	1.80	4.77	11.17
116.034	2.7E+04	0.29	116.03422	-0.22	2.5	C7 H10 O N	$[(M+H)-(C_2H_6O_3)-(H_2O)]^+$	1.80	4.77	11.17
126				-2.85	2.5	C7 H12 O2 N	$[(M+H)-(C_2H_6O_3)]^+$	1.80	4.77	11.17
147				-0.3	1.5	C8 H16 O3 N	$[(M+H)-(HCOOH)]^+$	1.80	4.77	11.17
182				0.26	3.5	C9 H14 O3 N	$[(M+H)-2(H_2O)]^+$	1.80	4.77	11.17
138				-0.46	2.5	C9 H16 O4 N	$[(M+H)-(H_2O)]^+$	1.80	4.77	11.17
Fragments ions and their isotopes										
Pseudo-molecular ion and its isotopes										
95	0.18	1.5	C9 H18 N O5			$[(M+H)]^+$		1.80	4.77	11.17
105	0.2	1.5	C8 13C H18 N O5			$[(M+H)]^+ (13C)$		1.80	4.77	11.17
94	-2.4	1.5	C9 H18 N O4 18O			$[(M+H)]^+ (18O)$		1.80	4.77	11.17
149	2.2	1.5	C8 13C H18 N O4 18O			$[(M+H)]^+ (13C+18O)$		1.80	4.77	11.17
238.12923	1.76E+03	0.02	238.128515	3	0.5	C9 H20 N O6	$[(M+H)+(H_2O)]^+$	1.80	4.77	11.17
242.09992	6.61E+06	68.15	242.099895	0.1	1.5	C9 H17 N Na O5	$[(M+Na)]^+$	1.80	4.77	11.17
243.10331	4.44E+05	4.58	243.10325	0.25	1.5	C8 13C H17 N Na O5	$[(M+Na)]^+ (13C)$	1.80	4.77	11.17
244.10363	1.79E+04	0.18	244.104139	-2.09	1.5	C9 H17 N Na O4 18O	$[(M+Na)]^+ (18O)$	1.80	4.77	11.17
258.07386	1.29E+05	1.33	258.073833	0.1	1.5	C9 H17 K N O5	$[(M+K)]^+$	1.80	4.77	11.17
259.07731	1.49E+03	0.02	259.077188	0.47	1.5	C8 13C H17 K N O5	$[(M+K)]^+ (13C)$	1.80	4.77	11.17
264.082	1.68E+05	1.74	264.08184	0.61	1.5	C9 H16 N Na2 O5	$[(M-H+2Na)]^+$	1.80	4.77	11.17
265.08498	3.08E+03	0.03	265.085195	-0.81	1.5	C8 13C H16 N Na2 O5	$[(M-H+2Na)]^+ (13C)$	1.80	4.77	11.17
310.08654	1.10E+03	0.01	310.08732	-2.52	1.5	C10 H18 N Na2 O7	$[(M+Na)+(HCOONa)]^+$	1.80	4.77	11.17
Adducts ions and their isotopes										
169	1.22	2.5	C18 H34 N2 Na O10			$[(2M+Na)]^+$		1.80	4.77	11.17
124	0.99	2.5	C17 13C H34 N2 Na O10			$[(2M+Na)]^+ (13C)$		1.80	4.77	11.17
179	0.26	2.5	C16 13C2 H34 N2 Na O10			$[(2M+Na)]^+ (13C2)$		1.80	4.77	11.17
114	0.1	2.5	C18 H33 N2 Na2 O10			$[(2M-H+2Na)]^+$		1.80	4.77	11.17

Reference compounds analysis :

Mass spectrum annotation

SPI-LIMS

Annotations **Annotation SPI**

1. FIA
2. PDBA
3. SPI

Polarité: Positive Négative

Décalage (ppm):

Précision (\pm ppm):

Temps de rétention:

Intervalle de confiance ($x \pm RT^y$): $x =$ $y =$

Colonnes chromatographiques:

FIA
 HPLC (C18) QTOF
 HSF5
 UPLC (C18)
 UPLC (C8)

Peaklists Fichier MS:

Outils

CEA - DSV/iBiTec-S/SPI

Reference compounds analysis: annotation using CEA spectral database

Pantothenic acid [M+H]⁺

M/Z	Intensity	Relative	M/Z (theo)	Delta (ppm)	RDBE	Composition	Attribution	FIA	UPLC (C8)	UPLC (C18)	HSF5
72.04431	7.7E+04	0.82	72.04439	-1.11	1.5	C3 H6 O N	$[(M+H)-(C_6H_{10}O_3)-(H_2O)]^+$	0	1.80	4.77	11.17
90.05482	3.6E+05	3.78	90.05496	-1.55	0.5	C3 H8 O2 N	$[(M+H)-(C_6H_{10}O_3)]^+$	0	1.80	4.77	11.17
116.034	2.7E+04	0.29	116.03422	-0.22	2.5	C7 H10 O N	$[(M+H)-(C_2H_6O_3)-(H_2O)]^+$	0	1.80	4.77	11.17
142.08585	1.0E+03	0.01	142.08626	-2.85	2.5	C7 H12 O2 N	$[(M+H)-(C_2H_6O_3)]^+$	0	1.80	4.77	11.17
174.11217	1.6E+04	0.17	174.11247	-0.3	1.5	C8 H16 O3 N	$[(M+H)-(HCOOH)]^+$	0	1.80	4.77	11.17
184.09687	2.4E+03	0.03	184.09682	0.26	3.5	C9 H14 O3 N	$[(M+H)-2(H_2O)]^+$	0	1.80	4.77	11.17
202.10729	1.5E+04	0.16	202.10738	-0.46	2.5	C9 H16 O4 N	$[(M+H)-(H_2O)]^+$	0	1.80	4.77	11.17
220.11799	9.69E+06	100	220.11795	0.18	1.5	C9 H18 N O5	$[(M+H)]^+$	0	1.80	4.77	11.17
221.12135	6.84E+05	7.06	221.121305	0.2	1.5	C8 13C H18 N O5	$[(M+H)]^+ (13C)$	0	1.80	4.77	11.17
222.12166	4.59E+04	0.47	222.122194	-2.4	1.5	C9 H18 N O4 18O	$[(M+H)]^+ (18O)$	0	1.80	4.77	11.17
223.12604	2.26E+03	0.02	223.125549	2.2	1.5	C8 13C H18 N O4 18O	$[(M+H)]^+ (13C+18O)$	0	1.80	4.77	11.17
238.12923	1.76E+03	0.02	238.128515	3	0.5	C9 H20 N O6	$[(M+H)+(H_2O)]^+$	0	1.80	4.77	11.17
242.09992	6.61E+06	68.15	242.099895	0.1	1.5	C9 H17 N Na O5	$[(M+Na)]^+$	0	1.80	4.77	11.17
243.10331	4.44E+05	4.58	243.10325	0.25	1.5	C8 13C H17 N Na O5	$[(M+Na)]^+ (13C)$	0	1.80	4.77	11.17
244.10363	1.79E+04	0.18	244.104139	-2.09	1.5	C9 H17 N Na O4 18O	$[(M+Na)]^+ (18O)$	0	1.80	4.77	11.17
258.07386	1.29E+05	1.33	258.073833	0.1	1.5	C9 H17 K N O5	$[(M+K)]^+$	0	1.80	4.77	11.17
259.07731	1.49E+03	0.02	259.077188	0.47	1.5	C8 13C H17 K N O5	$[(M+K)]^+ (13C)$	0	1.80	4.77	11.17
264.082	1.68E+05	1.74	264.08184	0.61	1.5	C9 H16 N Na2 O5	$[(M-H+2Na)]^+$	0	1.80	4.77	11.17
265.08498	3.08E+03	0.03	265.085195	-0.81	1.5	C8 13C H16 N Na2 O5	$[(M-H+2Na)]^+ (13C)$	0	1.80	4.77	11.17
310.08654	1.10E+03	0.01	310.08732	-2.52	1.5	C10 H18 N Na2 O7	$[(M+Na)+(HCOONa)]^+$	0	1.80	4.77	11.17
461.21113	6.45E+05	6.65	461.210569	1.22	2.5	C18 H34 N2 Na O10	$[(2M+Na)]^+$	0	1.80	4.77	11.17
462.21438	1.08E+05	1.11	462.213924	0.99	2.5	C17 13C H34 N2 Na O10	$[(2M+Na)]^+ (13C)$	0	1.80	4.77	11.17
463.2174	9.76E+03	0.1	463.217279	0.26	2.5	C16 13C2 H34 N2 Na O10	$[(2M+Na)]^+ (13C2)$	0	1.80	4.77	11.17
483.19256	4.21E+04	0.43	483.192514	0.1	2.5	C18 H33 N2 Na2 O10	$[(2M-H+2Na)]^+$	0	1.80	4.77	11.17



Human urines

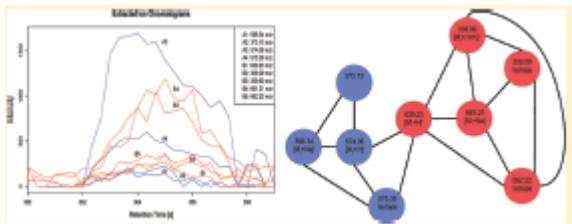
RT	MZ	#	Masse	Composé	Composition	Attribution	UPLC (C18)
4.72	202.107624	1	202.10738	pantothenic acid	C9 H16 O4 N	$[(M+H)-(H_2O)]^+$	4.77
4.73	90.054915	1	90.05496	pantothenic acid	C3 H8 O2 N	$[(M+H)-(C_6H_{10}O_3)]^+$	4.77
4.74	220.117542	1	220.11795	pantothenic acid	C9 H18 N O5	$[(M+H)]^+$	4.77
4.74	221.12168	1	221.121305	pantothenic acid	C8 13C H18 N O5	$[(M+H)]^+ (13C)$	4.77
4.74	222.122088	1	222.122194	pantothenic acid	C9 H18 N O4 18O	$[(M+H)]^+ (18O)$	4.77
4.74	242.100637	1	242.099895	pantothenic acid	C9 H17 N Na O5	$[(M+Na)]^+$	4.77
4.74	258.075314	1	258.073833	pantothenic acid	C9 H17 K N O5	$[(M+K)]^+$	4.77

Tools for annotation and metabolite identification, and data on biofluids begin to be published

analytical chemistry

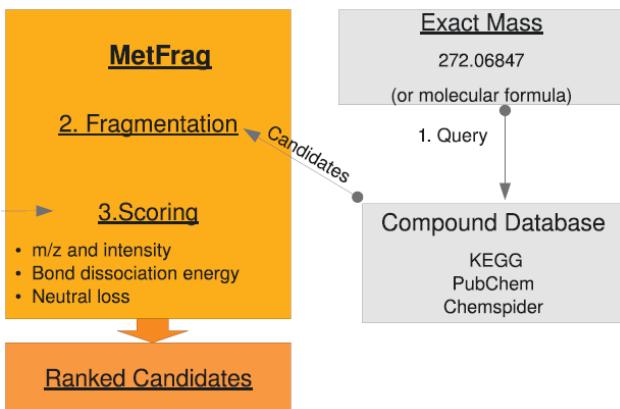
CAMERA: An Integrated Strategy for Compound Spectra Extraction and Annotation of Liquid Chromatography/Mass Spectrometry Data Sets

Carsten Kuhl,^{*†} Ralf Tautenhahn,[‡] Christoph Böttcher,[†] Tony R. Larson,[§] and Steffen Neumann^{*†}



In silico fragmentation for computer assisted identification of metabolite mass spectra

Sebastian Wolf^{1*}, Stephan Schmidt¹, Matthias Müller-Hannemann², Steffen Neumann¹



analytical chemistry

Annotation of the Human Adult Urinary Metabolome and Metabolite Identification Using Ultra High Performance Liquid Chromatography Coupled to a Linear Quadrupole Ion Trap-Orbitrap Mass Spectrometer

Aurelie Roux,[†] Ying Xu,[†] Jean-François Heilier,^{‡,§} Marie-Françoise Olivier,[†] Eric Ezan,[†] Jean-Claude Tabet,[‡] and Christophe Junot^{*†}

analytical chemistry

Evaluation of Coupling Reversed Phase, Aqueous Normal Phase, and Hydrophilic Interaction Liquid Chromatography with Orbitrap Mass Spectrometry for Metabolomic Studies of Human Urine

Tong Zhang,^{*†} Darren J. Creek,^{‡,§} Michael P. Barrett,[‡] Gavin Blackburn,[†] and David G. Watson[†]

Journal of proteome research

The Footprints of Gut Microbial–Mammalian Co-Metabolism

Xiaojiao Zheng,^{†,‡} Guoxiang Xie,[†] Aihua Zhao,[†] Linjing Zhao,[†] Chun Yao,[†] Norman H. L. Chiu,[§] Zhanxiang Zhou,[†] Yuqian Bao,[¶] Weiping Jia,^{¶,||} Jeremy K. Nicholson,^{||} and Wei Jia^{*†}

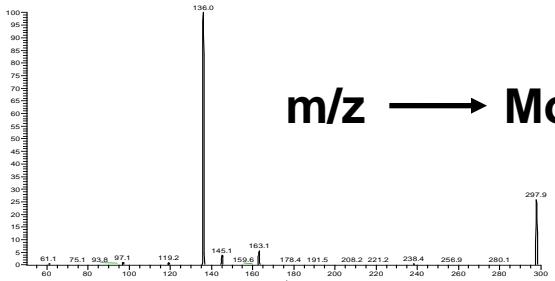
HMDB 3.0—The Human Metabolome Database in 2013

Nucleic Acids Research, 2013, Vol. 41, Database issue D801-D807
doi:10.1093/nar/gks1065

Biochemical data on ~ 40000 metabolites
Spectral data (RMN, MS)

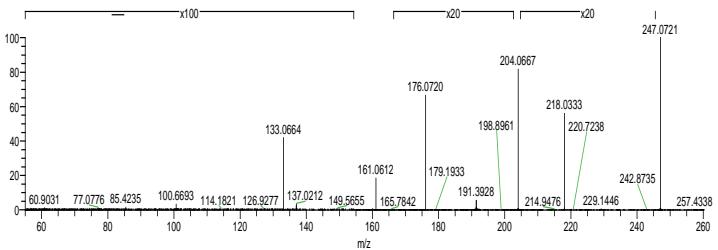
Formal identification of metabolites often requires several complementary analytical tools

Mass spectrometry



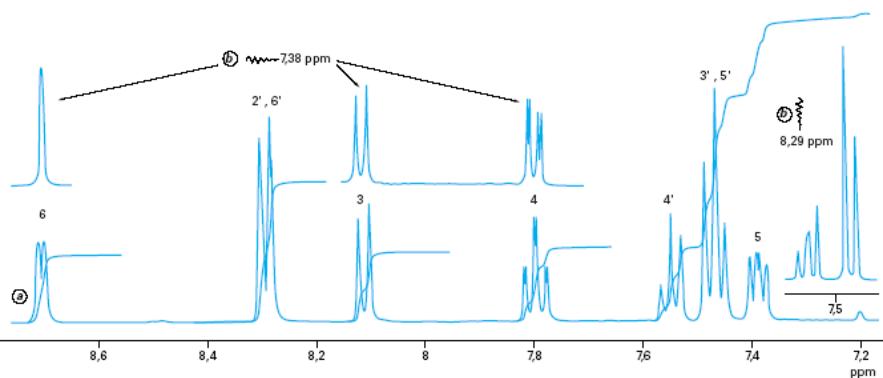
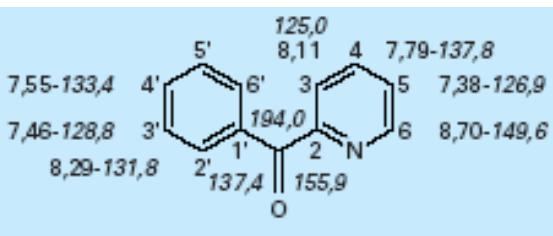
m/z → Molecular mass

MS/MS, MSⁿ



Structural information

NMR spectroscopy



To discriminate between isomers

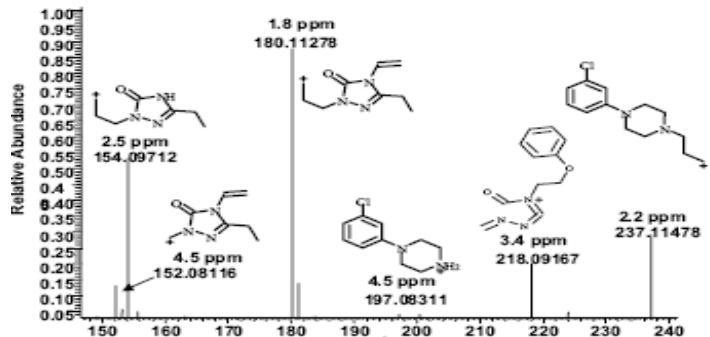
L'identification des métabolites:

1. Spectre de masse

- Isoler $[M+H]^+$ ou $[M-H]^-$
- composition élémentaire ($C_xH_yO_z$)
- Isotopes (^{13}C , ^{18}O , ^{34}S ...)

Annotation

2. Bases de données



3. Spectres de fragmentation (CID)

4. Expériences complémentaires (échange H/D, RMN)

5. Confirmation

- Synthèse chimique
- Analyse LC/MS comparative

Metabolite identification

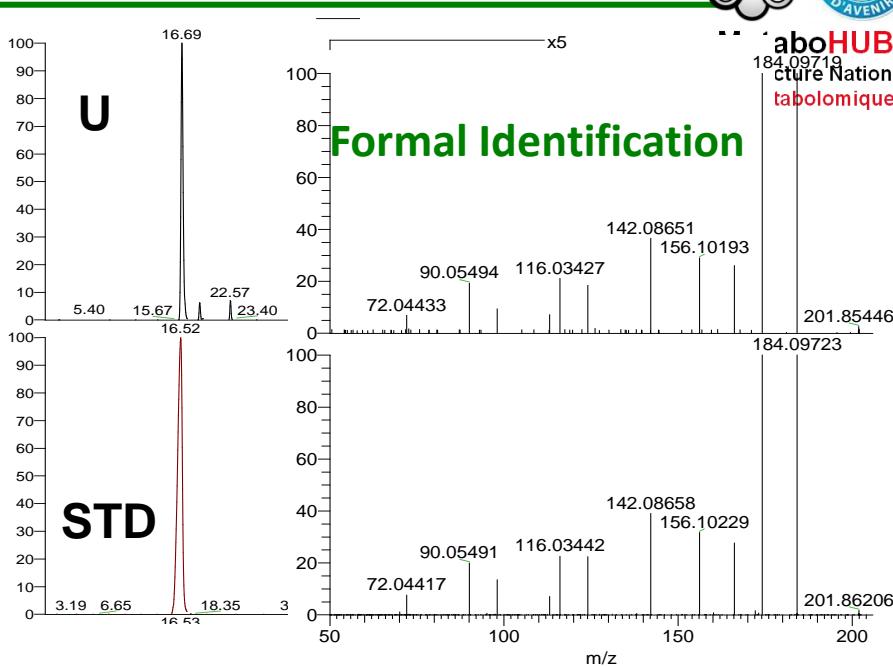
Metabolomics (2007) 3:211–221
DOI 10.1007/s11306-007-0082-2

ORIGINAL ARTICLE

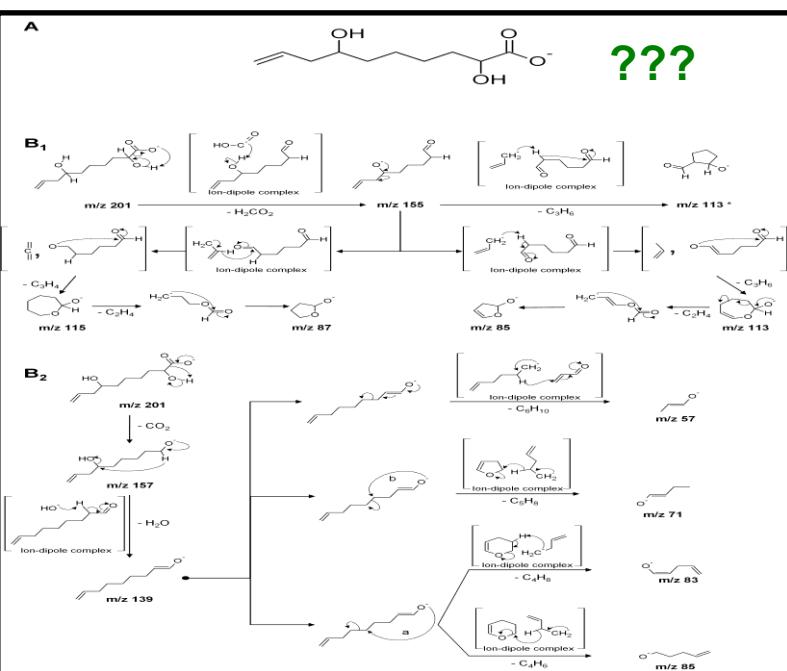
Proposed minimum reporting standards for chemical analysis

Chemical Analysis Working Group (CAWG) Metabolomics Standards Initiative (MSI)

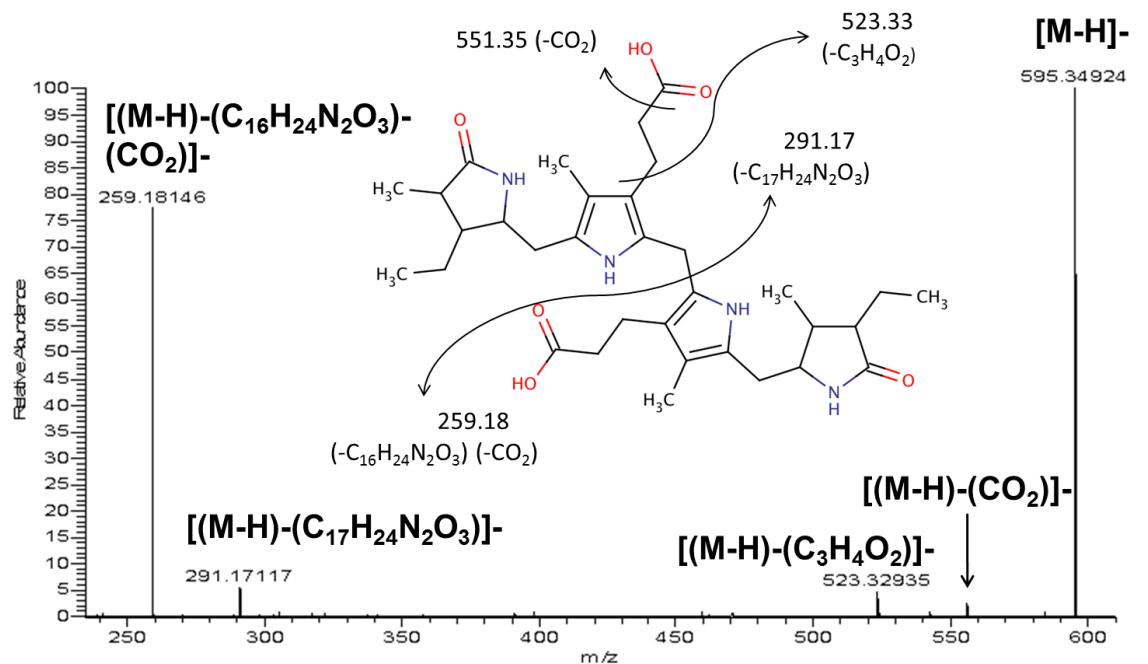
Lloyd W. Sumner · Alexander Amberg · Dave Barrett · Michael H. Beale ·
Richard Beger · Clare A. Daykin · Teresa W.-M. Fan · Oliver Fiehn ·
Royston Goodacre · Julian L. Griffin · Thomas Hankemeier · Nigel Hardy ·
James Harnly · Richard Higashi · Joachim Kopka · Andrew N. Lane ·
John C. Lindon · Philip Marriott · Andrew W. Nicholls · Michael D. Reily ·
John J. Thaden · Mark R. Viant



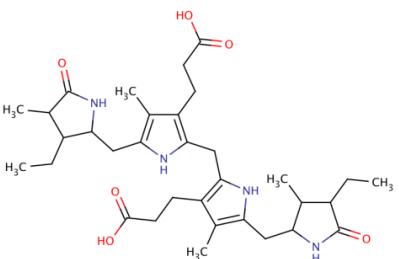
- Identified compounds (see below).
- Putatively annotated compounds (e.g. without chemical reference standards, based upon physicochemical properties and/or spectral similarity with public/commercial spectral libraries).
- Putatively characterized compound classes (e.g. based upon characteristic physicochemical properties of a chemical class of compounds, or by spectral similarity to known compounds of a chemical class).
- Unknown compounds—although unidentified or unclassified these metabolites can still be differentiated and quantified based upon spectral data.



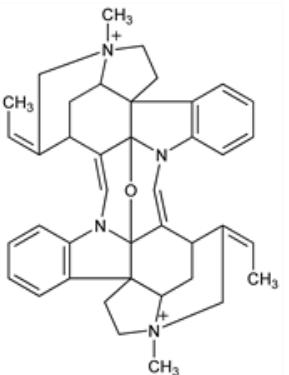
Variable number	XCMS output		CAMERA output			Inter-sample correlation	Public database annotation
	m/z	Retention time (min)	isotopes	adduct	pcgroup		
1806	303.1443	9.33	**	**	531	NA	**
4663	593.3312	9.34	[681][M]+	**	512	NA	L-Urobilin
4668	594.3368	9.34	[681][M+1]+	**	512	NA	**
4679	595.3463	9.40	[650][M]+	[M-H]-	394	1.00	C-Curarine / L-Urobilinogen
4682	596.3514	9.40	[650][M+1]+	**	394	0.98	**
4878	631.3256	9.40	**	[M+Cl]-	394	0.96	**
3797	481.2789	9.46	**	**	552	NA	GPCho(10:0/4:0) / GPCho(12:0/2:0)
2763	381.1910	9.53	**	**	627	NA	**
3834	485.1792	9.61	**	**	544	NA	Rutaevin / Nafenopinglucuronide
1255	253.1440	9.67	**	**	556	NA	**



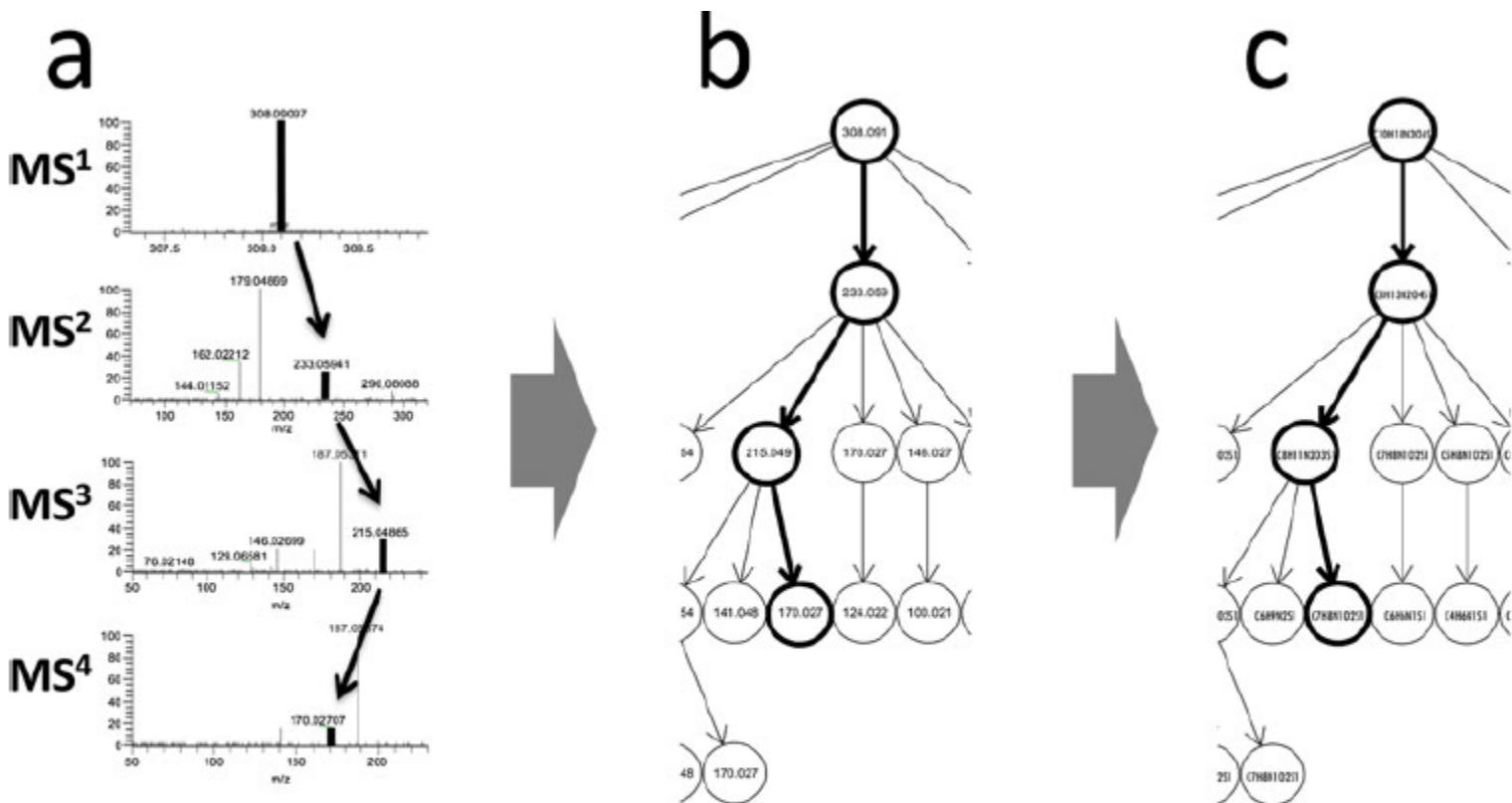
L-Urobilinogen



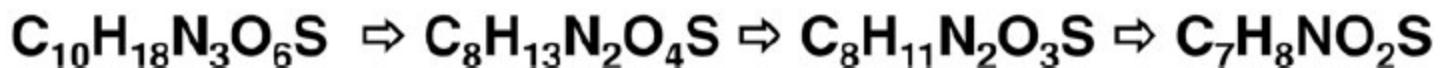
C-Curarine



Automated analysis of multistage MS (MS^n): Spectral trees



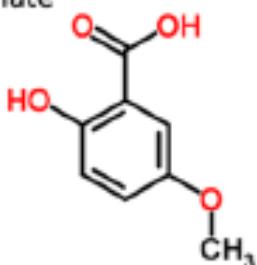
d



(Kasper PT, Rapid Commun. Mass Spectrom., 2012)

Automated analysis of multistage MS (MS^n): Spectral trees

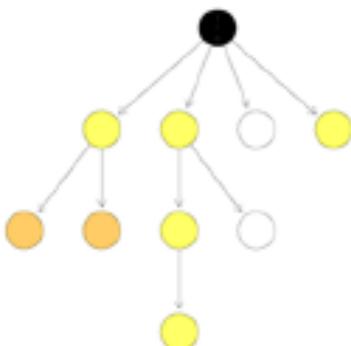
Id_7: 5-Methoxysalicylate



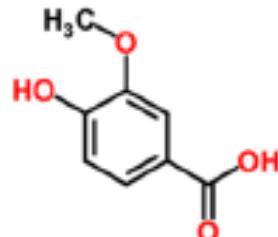
Tree of fragments



Tree of losses



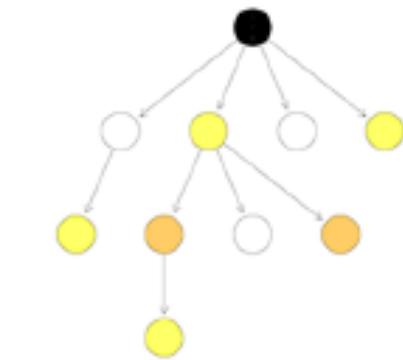
Id_8: Vanillic acid



Tree of fragments

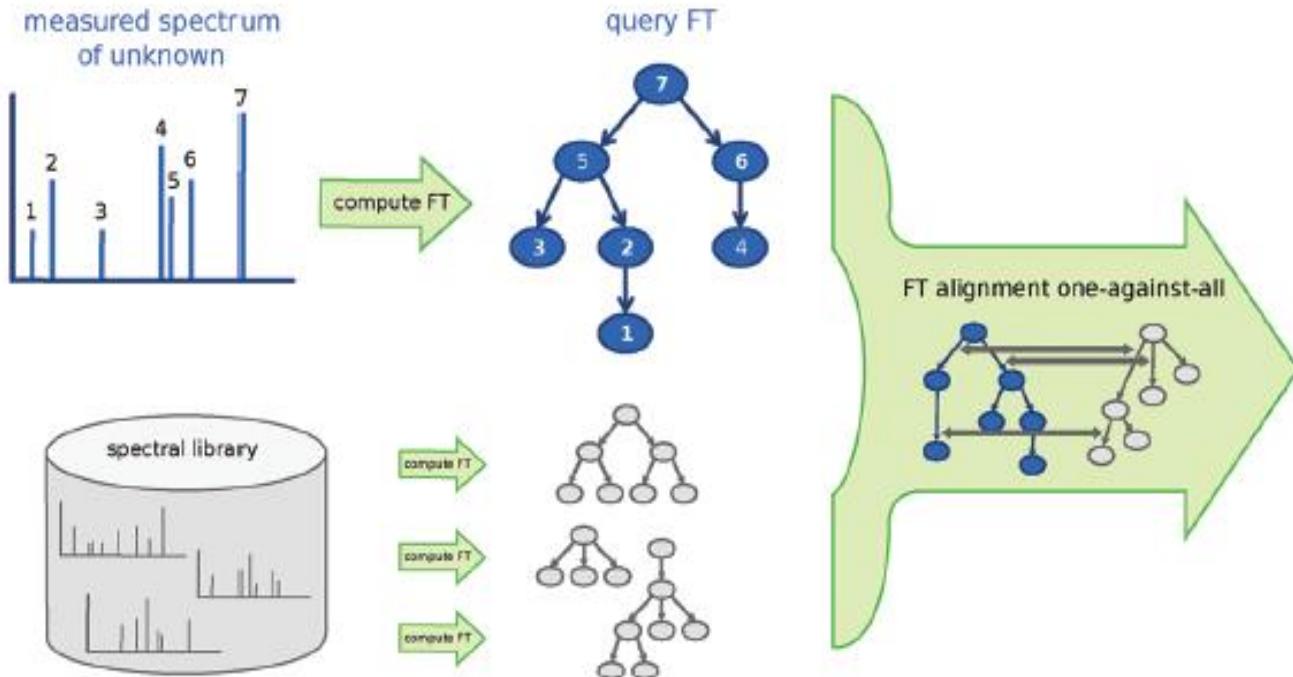


Tree of losses

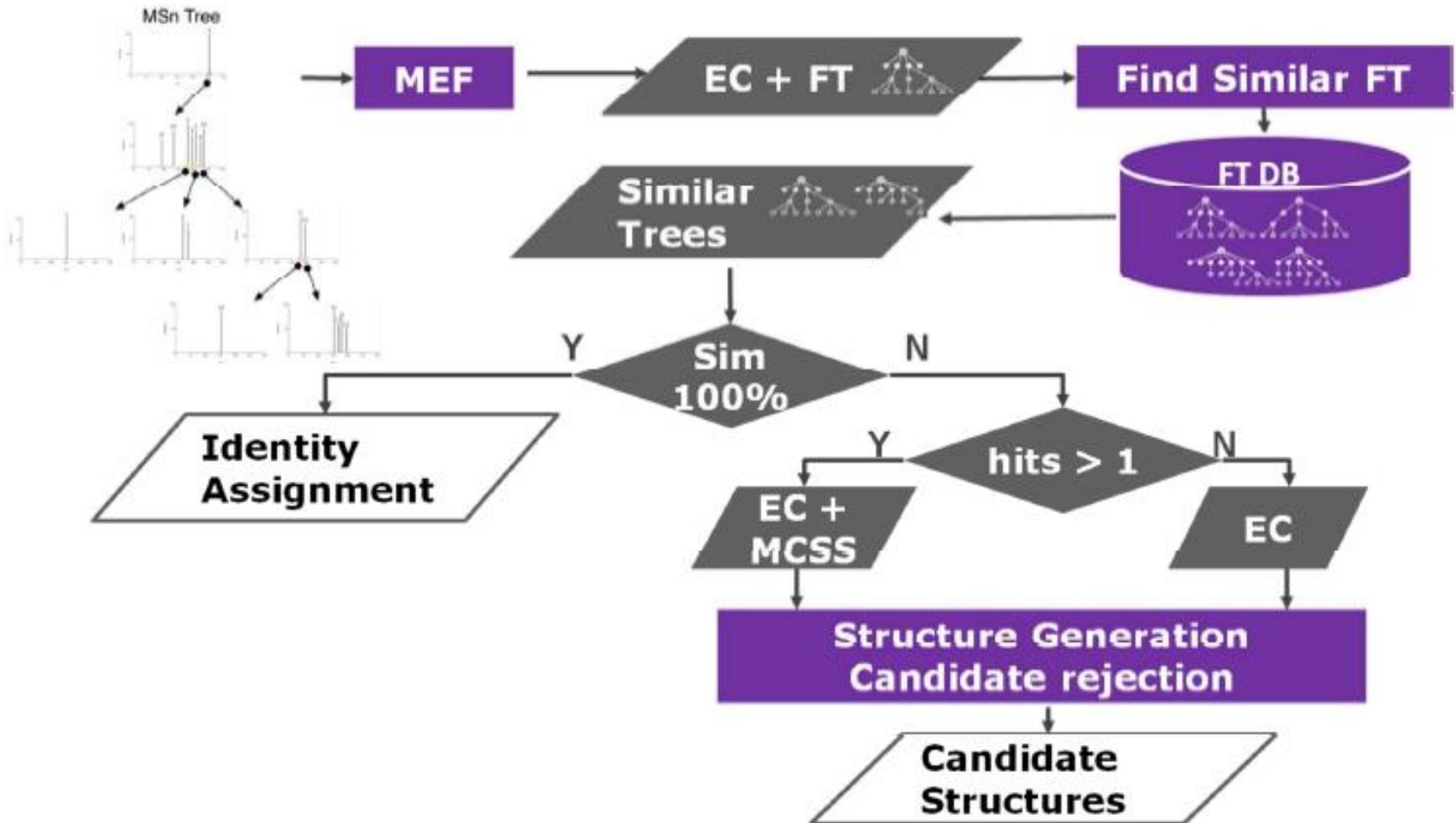


Automated analysis of multistage MS (MS^n): Spectral trees

(c) FT-BLAST: searching in reference database



Automated analysis of multistage MS (MS^n): Spectral trees



Mise en place d'une méthode d'étalonnage pour la construction d'une base de données MS/MS en science métabolomique

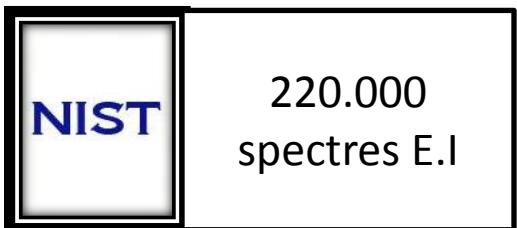
F. Ichou, D. Lesage, C. Junot, J-C. Tabet



Travail actuellement poursuivi et coordonné par **R. Cole** dans le cadre de l'infrastructure MetaboHUB

Les bases de données

- Majoritairement en ionisation par électron (E.I)
 - ✓ En E.I : beaucoup de fragmentations, absence parfois de l'ion moléculaire M^{+} et spectres de masse reproductibles
 - ✓ Nombreuses bases de données et de tailles importantes
 - ✓ Exemple :



- En API : Peu de fragmentations, peu reproductible et interférence de la matrice
 - Conséquence : bases de données MS peu fiables
 - Généralement bases de données MS/MS
 - ✓ Problème de reproductibilité en CID à un régime de basse énergie
 - ✓ 2 approches utilisées

Les différentes approches de base de données MS/MS:

1 - Approche non-standardisée

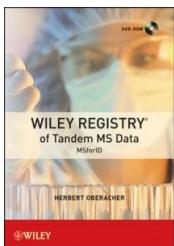
2 - Approche standardisée

Principe reposant sur l'enregistrement de multiples empreintes CID



■ *H.Oberacher et M. Pavlic*⁷

- ✓ Analyses à 10 énergies de collision
- ✓ Elimination de l'ion parent
 - Grande variation de l'abondance
- ✓ Amélioration de l'algorithme de matching

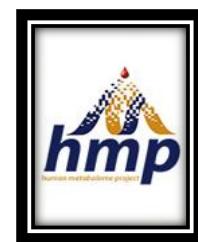


'MSFor ID library'

✓ Autres exemples :



Test



Les différentes approches de bases de données MS/MS:

1 - Approche non-standardisée

2 - Approche standardisée

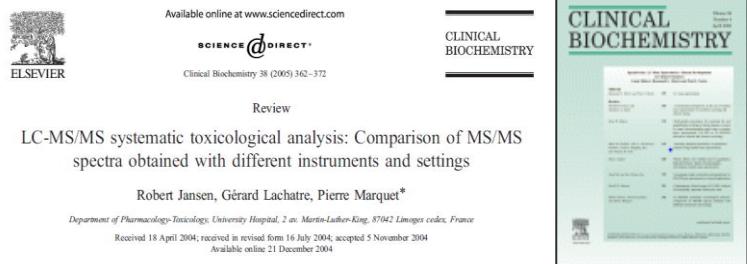
Etalonnage avec un composé

■ **P. Marquet et al⁸**

✓ Glafenine

■ **C.Hopley and T. Bristow⁹**

✓ Reserpine



Available online at www.sciencedirect.com
SCIENCE @ DIRECT®
Clinical Biochemistry 38 (2005) 362–372
Review
LC-MS/MS systematic toxicological analysis: Comparison of MS/MS spectra obtained with different instruments and settings
 Robert Jansen, Gérard Lachatre, Pierre Marquet*
Department of Pharmacology-Toxicology, University Hospital, 2 av. Martin-Luther-King, 87042 Limoges cedex, France
 Received 18 April 2004; revised 16 July 2004; accepted 5 November 2004
 Available online 21 December 2004

RAPID COMMUNICATIONS IN MASS SPECTROMETRY
Rapid Commun. Mass Spectrom. 2008, 22: 1779–1786
 Published online in Wiley InterScience (www.interscience.wiley.com) DOI: 10.1002/rcm.3545

Towards a universal product ion mass spectral library – reproducibility of product ion spectra across eleven different mass spectrometers

Chris Hopley^{1*}, Tony Bristow², Anneke Lubben³, Alec Simpson⁴, Elaine Bull⁴,
 Katerina Klagkou⁵, Julie Herniman⁶ and John Langley⁶

¹LGC Queens Road, Teddington TW11 0LX, UK

²AstraZeneca, Macclesfield SK10 2NA, UK

³Department of Chemistry, University of Cambridge, Cambridge CB3 0FW, UK

⁴QeumSoft Inc., Burnham-on-Sea, Somerset BS11 9NY, UK

⁵ThermoElectron, Stafford House, Boundary Way, Hemel Hempstead HP2 7GE, UK

⁶School of Chemistry, University of Southampton, Southampton SO17 1BJ, UK

Received 8 October 2007; Revised 20 March 2008; Accepted 20 March 2008

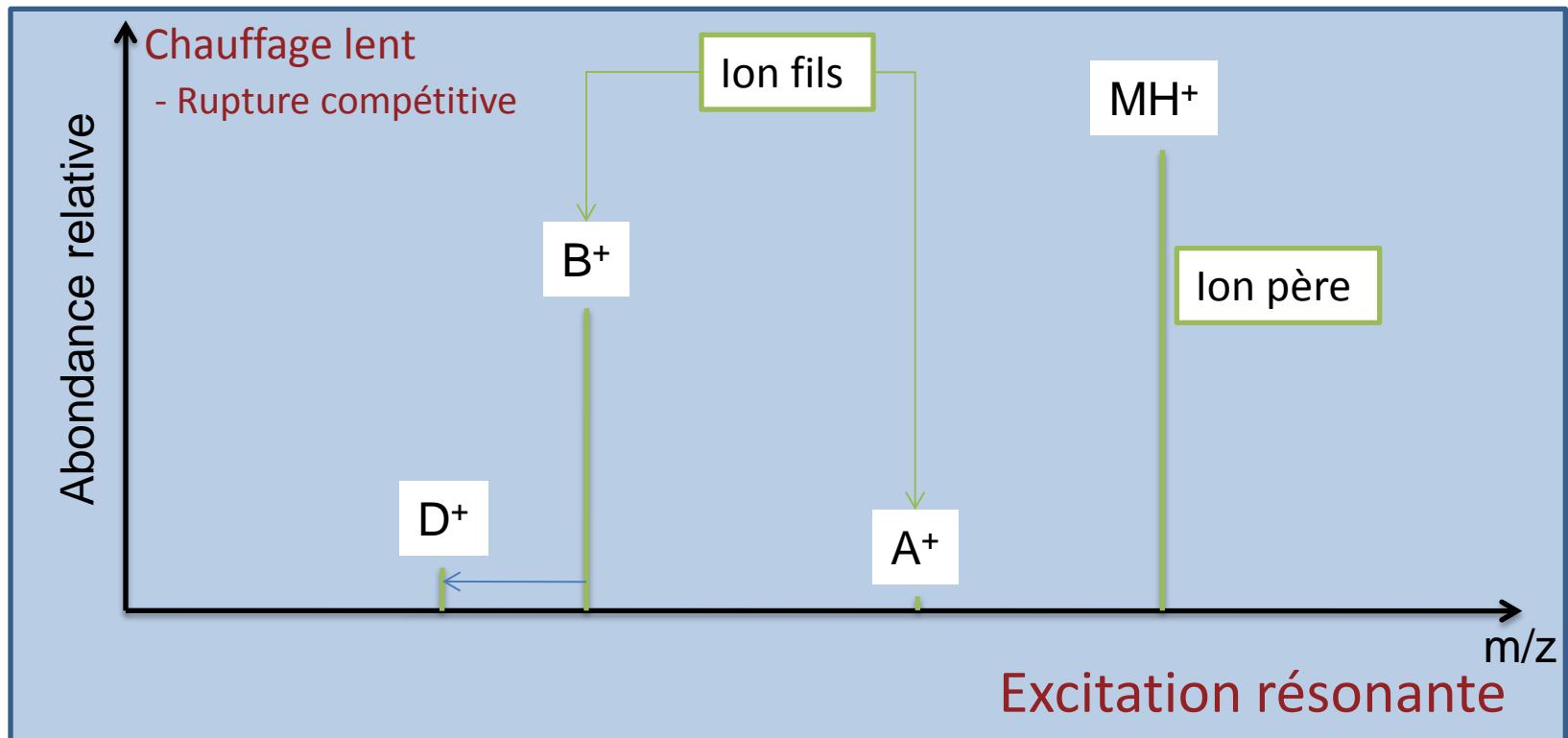


⁸P. Marquet et al, Clin. biochem. 2005, 38, 362

⁹C.Hopley, T. Bristow, Rap. Com. Mass Spectrom. 2008, 22, 1779

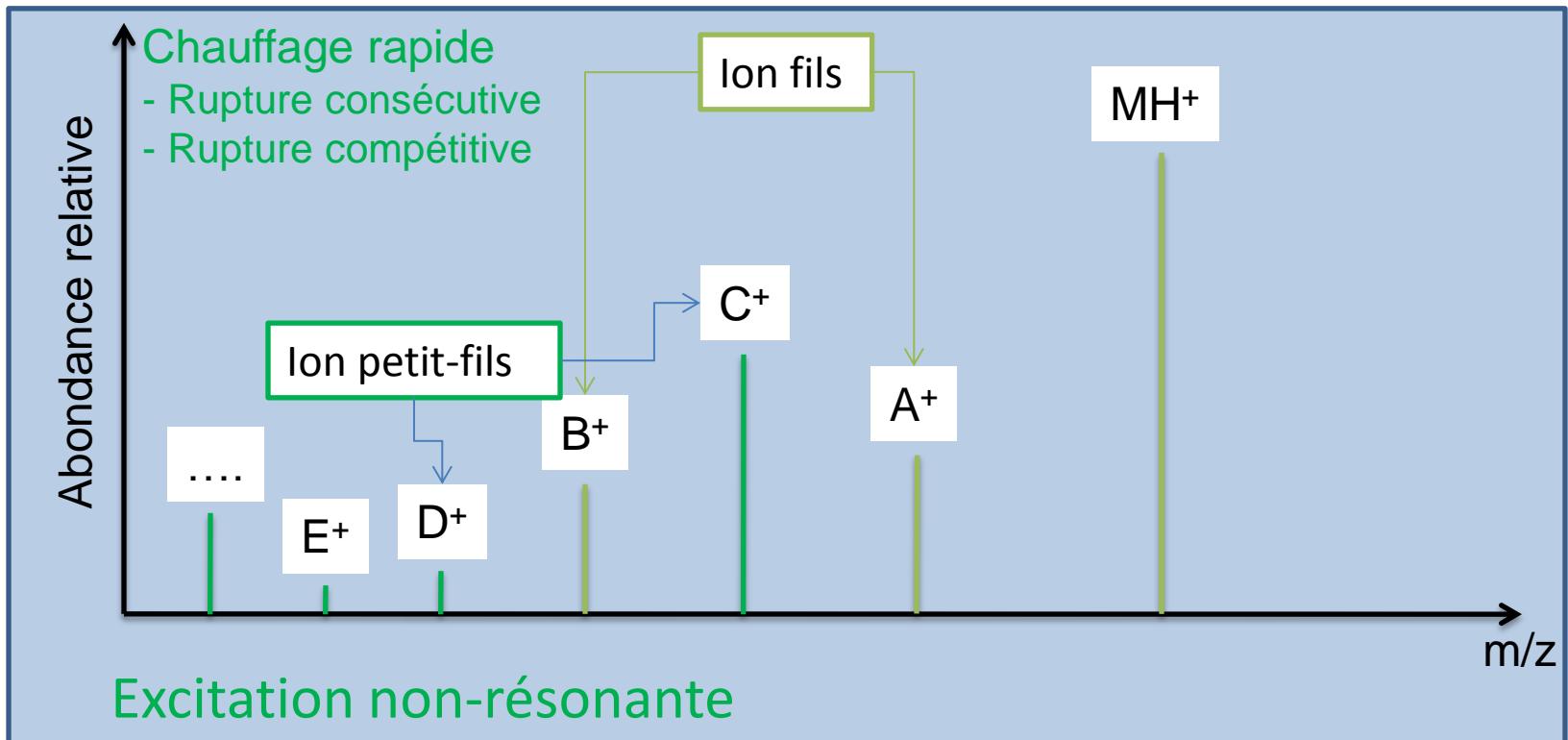
Type d'excitation

Instruments à piégeage d'ions (Piège LTQ, 3D)



Type d'excitation

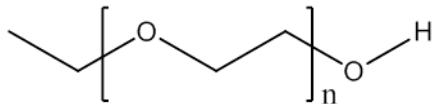
Instruments à faisceau d'ions (TQ, QTOF,...)



Procédure d'étalonnage

- Ajuster les ratios des abondances relatives des ions produits par la fragmentation du Polyéthylène glycol (PEG) pour avoir des spectres CID comparables à un spectre de référence

Polyethylene Glycol (PEG_{200/300})



- Instruments à piégeage d'ions (Résonant) :
 - ✓ Choisir la même énergie de collision normalisée (15%, 20% et 25%)
 - ✓ Ajuster le temps d'activation
- Instruments à faisceau d'ions (Non-résonant) :
 - ✓ Choisir trois énergies de collision
 - ✓ Ajuster la pression de la cellule de collision

Objectifs

Evaluation de la procédure :

Reproductibilité inter-laboratoire sur le même type d'instruments

- ✓ Est-il possible d'utiliser des spectres de référence enregistrés dans d'autres laboratoires pour identifier les métabolites ?

Comparaison inter-instrumentale

- ✓ Les spectres CID provenant de différents instruments donnent-ils les mêmes informations ?

Quelles sont les limites de la méthode d'étalonnage ?

- ✓ Doit-on construire une base de données MS/MS spécifique à chaque type d'instruments (tandems à faisceau d'ions ou à piégeage d'ions) ?

Validation de la procédure

- 19 composés :
 - ✓ Panel représentatif de la diversité des problèmes rencontrés en métabolomique
- 9 laboratoires partenaires
- 13 instruments (QTOF, LTQ, Orbitrap)
- Mise en situation avec un logiciel commercial (SmileMS)

Panel de composés

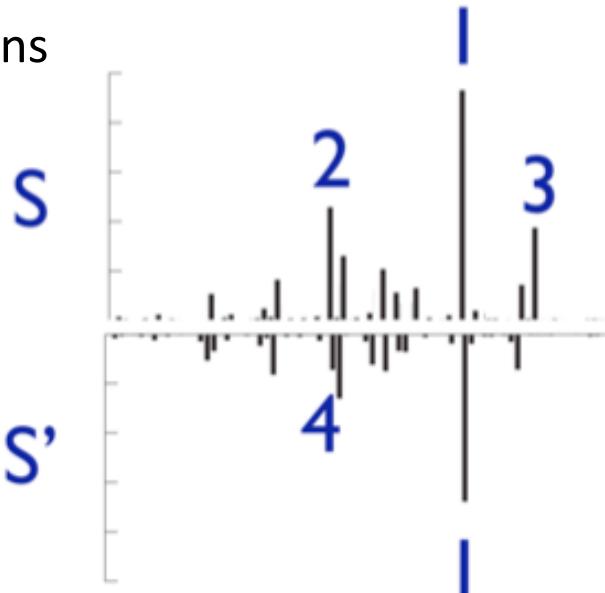
Composé	Masse molaire (g/mol)	Polarité	Caractéristique des composés
Caféine	194	Pos	
Myricetine	318	Pos&Neg	Groupe A.
Xanthosine	284	Pos&Neg	Peu de voies de dissociations compétitives (< 2 ions fragments dans les pièges à ions)
Inosine	268	Pos&Neg	
5-Methoxyindoleacetate	205	Pos	
Acide Indoleacrylique	187	Neg	
Acide 12-hydroxydodecanoïque	216	Neg	
Acide Xanthurenique	205	Neg	Groupe B.
Dimethoate	229	Pos	2-3 ions fragments
Naringénine	272	Pos	
Tyrosine	181	Pos&Neg	
Panthenol	205	Pos&Neg	
Acide Cis-aconitique	174	Neg	
Acide Trans-aconitique	174	Neg	
Cystathionine	222	Pos&Neg	Groupe C.
Acetamiprid	222	Pos&Neg	Beaucoup de voies de dissociations (> 3 ions fragments)
Arginine	174	Pos&Neg	
Glutathion	307	Pos&Neg	
Carnosine	226	Pos&Neg	

Logiciel SmileMS

➤ Originalité de l'algorithme X-Rank

- Comparaison par rapport aux m/z et indirectement aux abondances relatives des ions
- Classification des fragments par ordre d'intensité des pics
- Compatibilité Inter-instrumentale
 - Extraction par protéowizards des différents formats constructeurs

Conversion sous .mzXML, .mgf



Principe de l'expérience

Comparaison :

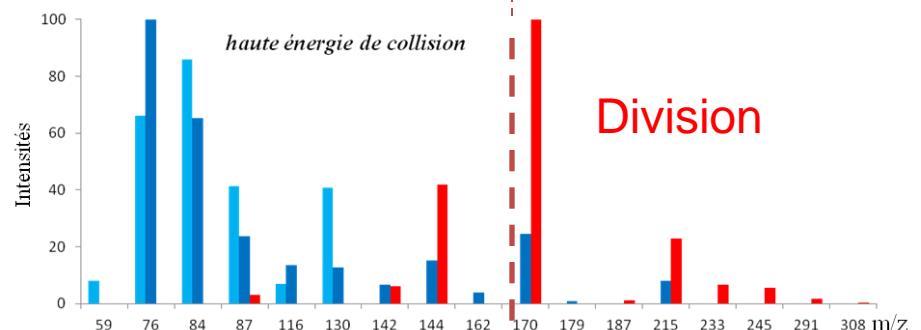
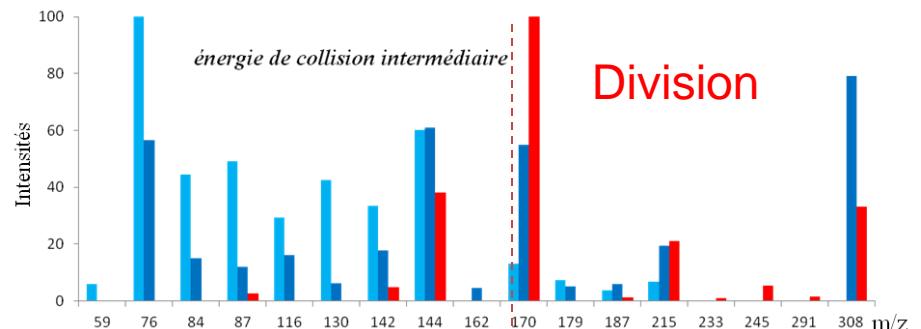
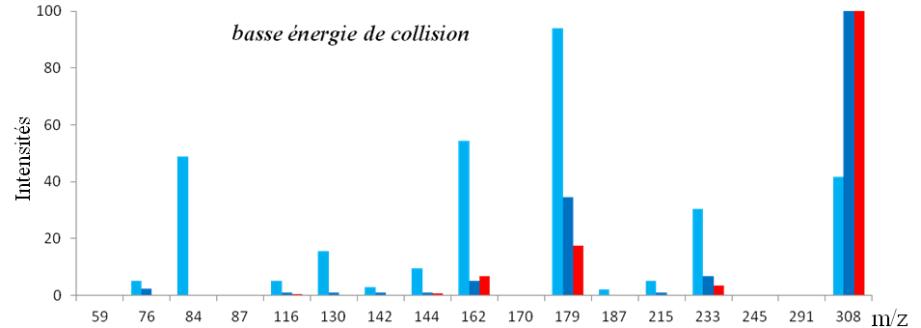
- ✓ Inter-QTOF
- ✓ Inter-LTQ
- ✓ Inter-Orbitrap
- ✓ Inter-instrumentale

Deux principaux résultats :

- Score d'identification
 - ✓ Score dépendant de la quantité d'informations contenues dans le spectre CID
 - ✓ Nette baisse du score en négatif sauf pour les molécules ayant beaucoup d'ions fragments
 - Mode négatif plus pauvre en fragments
- Ecart-type

Résultats inter-instrumentaux

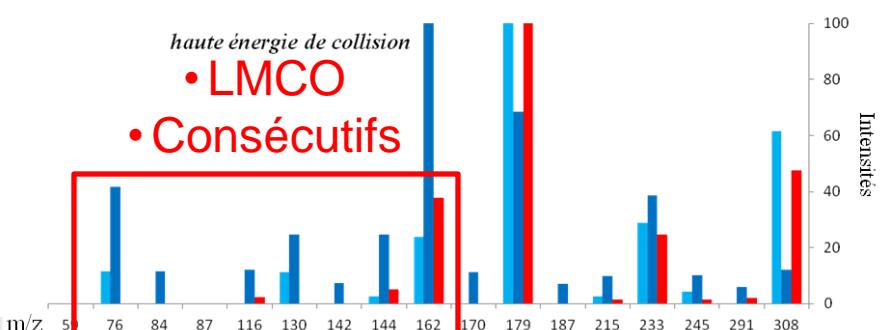
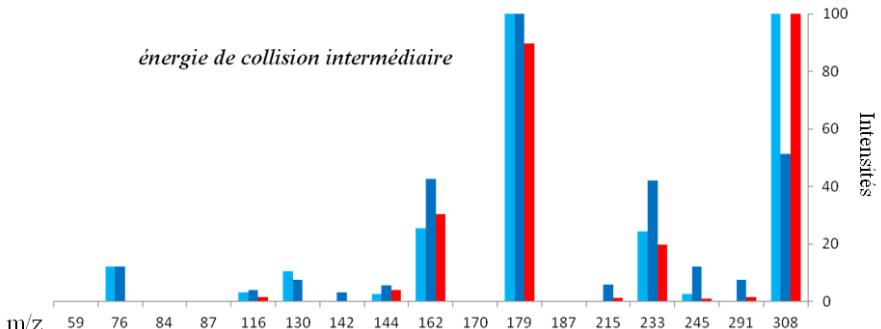
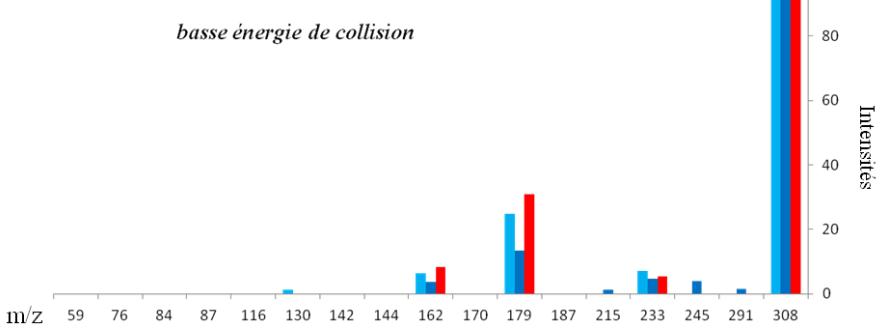
Avant standardisation



■ Orbitrap

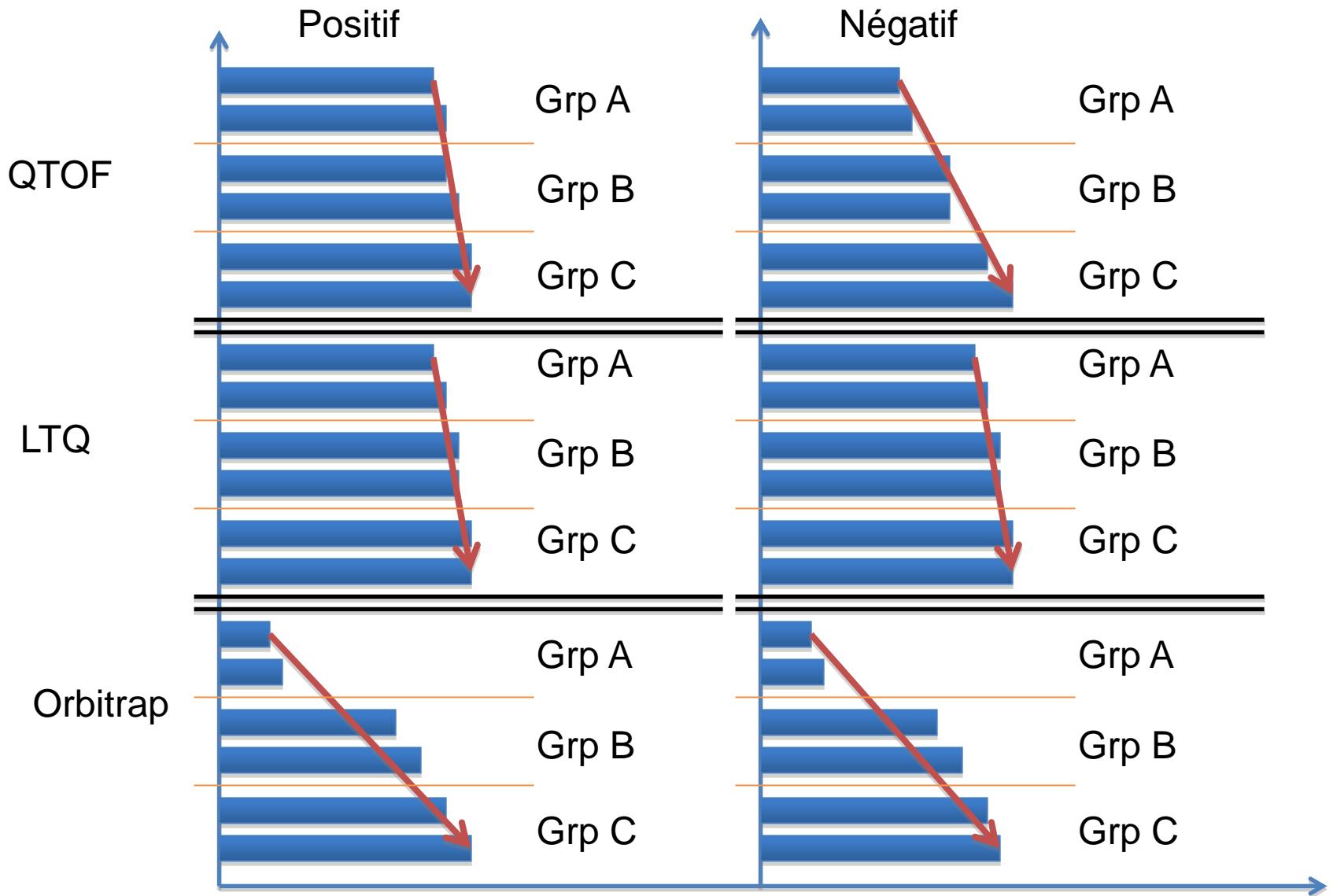
■ TQ

Après standardisation



■ QTOF

Quantité d'informations

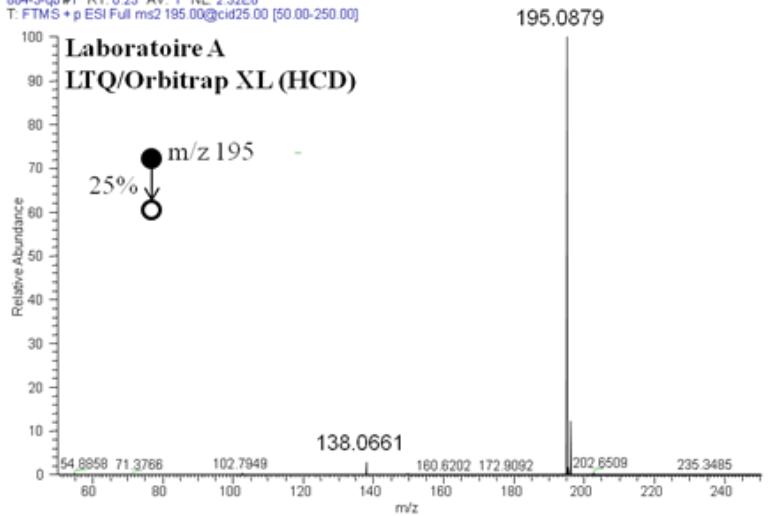


Résultats inter-Orbitrap

Score moyen d'identification et écart-types proches de 0

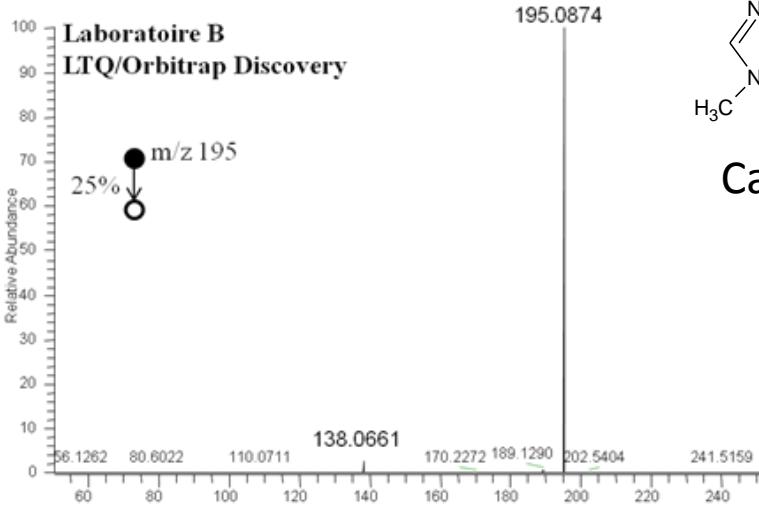
a) 084-3-qb#1 RT: 0.23 AV: 1 NL: 2.32E6
T: FTMS + p ESI Full ms2 195.00@cid25.00 [50.00-250.00]

Laboratoire A
LTQ/Orbitrap XL (HCD)



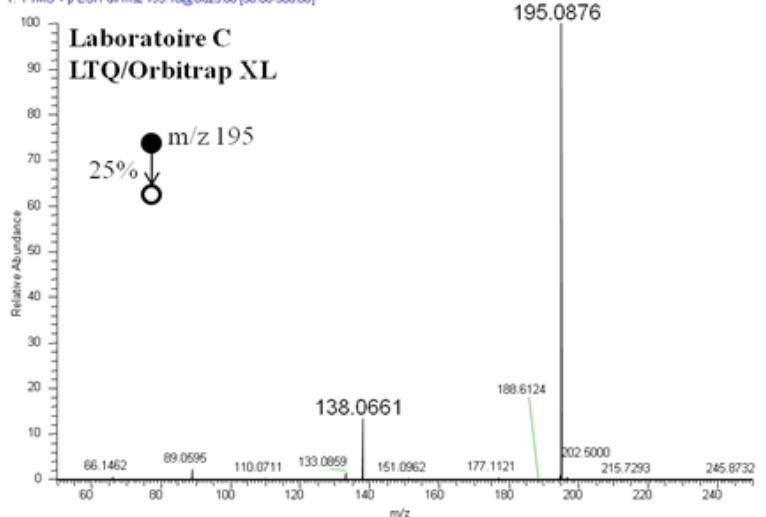
b) 20111118-PO5-011-3-qb#1 RT: 0.56 AV: 1 NL: 1.88E7
T: FTMS + p ESI Full ms2 195.08@cid25.00 [50.00-250.00]

Laboratoire B
LTQ/Orbitrap Discovery



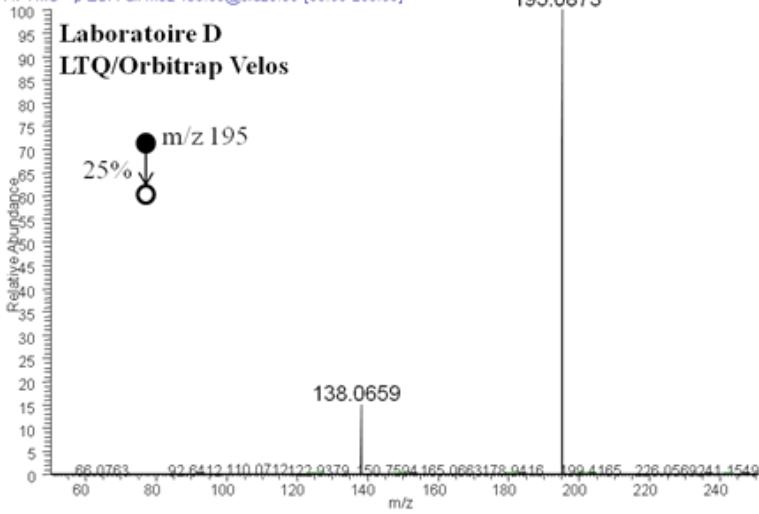
c) 2011-0811060 #27-52 RT: 0.54-0.78 AV: 26 NL: 9.42E6
T: FTMS + p ESI Full ms2 195.10@cid25.00 [50.00-500.00]

Laboratoire C
LTQ/Orbitrap XL

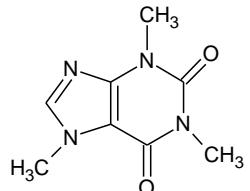


d) caffeine_Orbitrap_CES5-84 RT: 0.46-0.70 AV: 32 NL: 2.56E7
T: FTMS + p ESI Full ms2 195.00@cid25.00 [50.00-250.00]

Laboratoire D
LTQ/Orbitrap Velos



En mode positif



Caféine

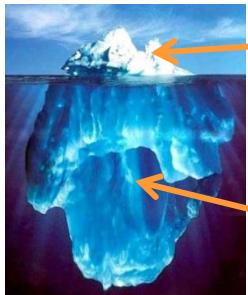
Conclusion

➤ **Procédure de standardisation**

- Meilleure reproductibilité inter-laboratoires et inter-instrumentale
- Intérêts de la procédure d'étalonnage
 - ✓ Contrôle de l'énergie interne
 - ✓ Réduction des ruptures consécutives pour les tandems à faisceau d'ions
 - ✓ Chaque instrument peut être une référence
- 2 Sources de déviation dans les scores
 - ✓ Les molécules avec peu de fragmentations
 - ✓ Les tandems à faisceau d'ions (contrôle énergétique plus difficile)

➤ **Perspectives**

- ✓ Validation et construction de la librairie
- ✓ Projet MétaboHUB



Standards commerciaux

Composés issus de matrices biologiques,...

Several methods have to be implemented in order to achieve an optimal metabolome coverage

[dx.doi.org/10.1021/ac2030738](https://doi.org/10.1021/ac2030738) | *Anal. Chem.* 2012, 84, 1994–2001

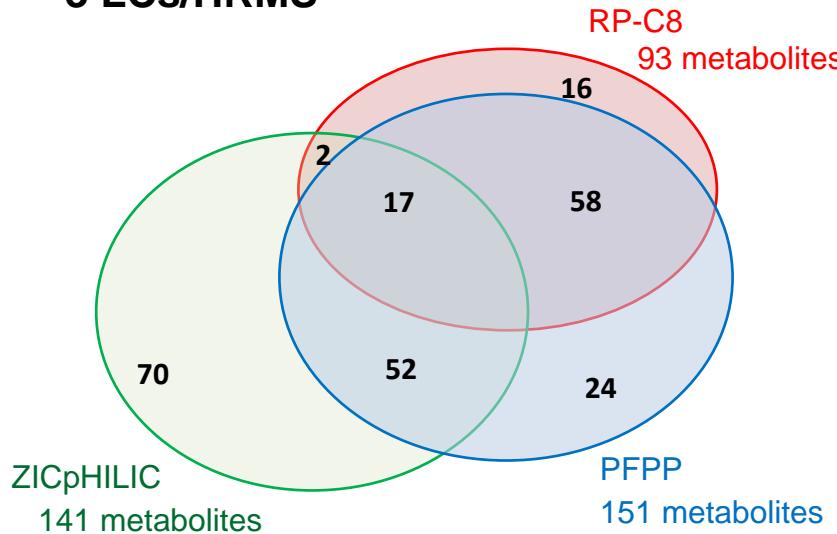
Evaluation of Coupling Reversed Phase, Aqueous Normal Phase, and Hydrophilic Interaction Liquid Chromatography with Orbitrap Mass Spectrometry for Metabolomic Studies of Human Urine

Tong Zhang,^{*,†} Darren J. Creek,^{‡,§} Michael P. Barrett,[‡] Gavin Blackburn,[†] and David G. Watson[†]

[dx.doi.org/10.1021/ac300586m](https://doi.org/10.1021/ac300586m) | *Anal. Chem.* 2012, 84, 6963–6972

270 metabolites identified in human plasma

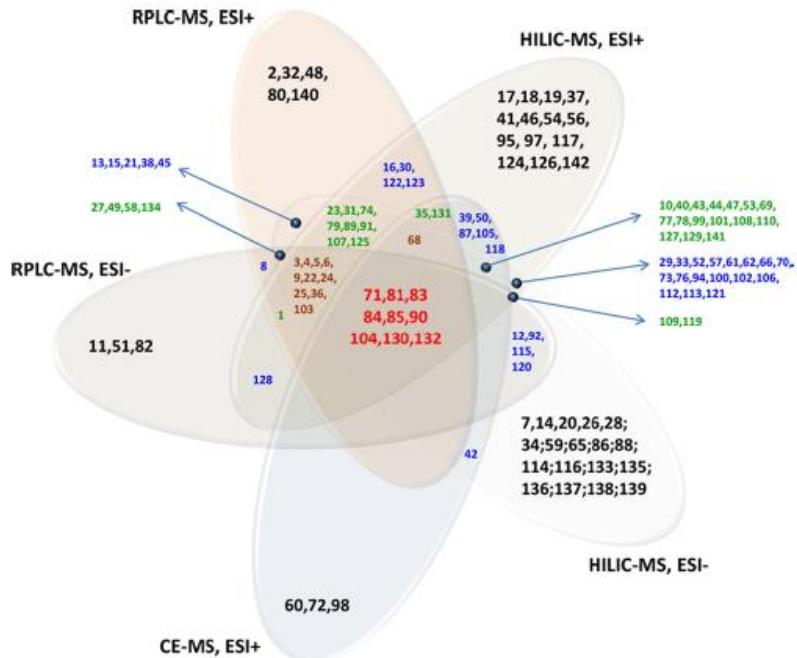
3 LCs/HRMS



(Boudah et al., *J. Chromatogr. B*, 2014)

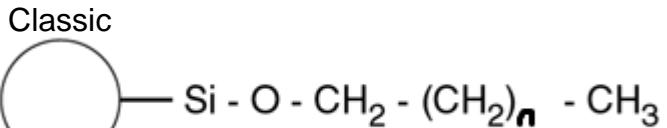
Systematic Evaluation of Extraction Methods for Multiplatform-Based Metabotyping: Application to the *Fasciola hepatica* Metabolome

Jasmina Saric,[†] Elizabeth J. Want,[†] Urs Duthaler,[‡] Matthew Lewis,[†] Jennifer Keiser,[‡] John P. Shokcor,[§] Gordon A. Ross,[#] Jeremy K. Nicholson,[†] Elaine Holmes,^{*,†} and Marina F. M. Tavares^{*,†,||}

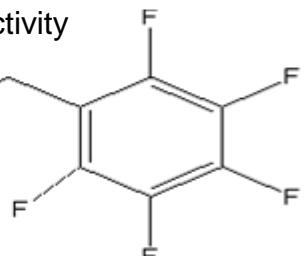


The is a need to add other separative dimensions to HRMS

Reverse Phase Liquid Chromatography



C8,
C18...
.... Alternative selectivity



PFPP

PentaFluoroPhenylPropyl bound
silica

Retention mechanism :

- **Hydrophobic interactions**

Bile acids, Steroid hormones...

Retention mechanism :

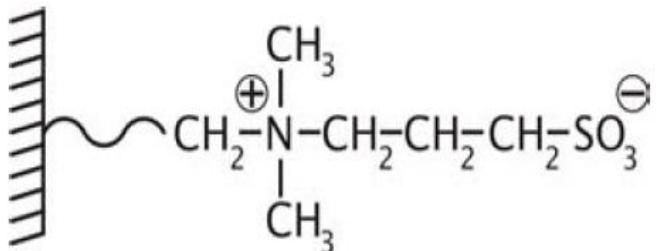
- **$\pi-\pi$ interactions** for phenyl-based compounds
- **dipole-dipole** interactions for halo/polar compounds

Amino acids , cyclic compounds...

Most widely used columns:

- aqueous and organic solvents compatibility
- robustness
- well defined retention mechanisms
- continuous manufacturer improvement

Hydrophilic Interaction Liquid Chromatography



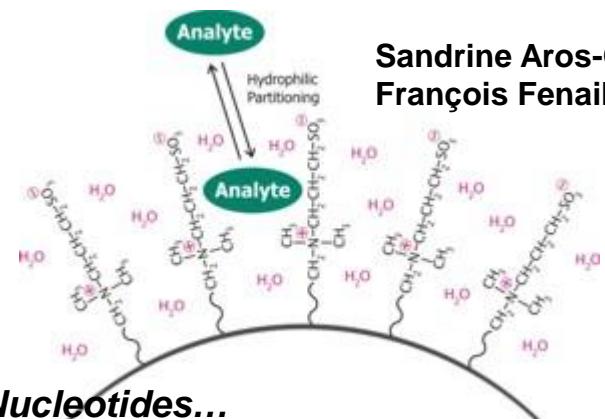
ZIC HILIC

Zwitterionic sulfobetaine bound
silica

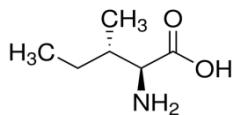
Retention mechanism :

- **hydrophilic partitioning** from the eluent to the enriched-water layer
- **electrostatic interactions** with either positive and negative charges

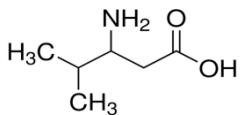
Carboxylic acids, Sugars, Nucleotides...



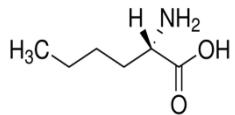
To discriminate between
isomer species



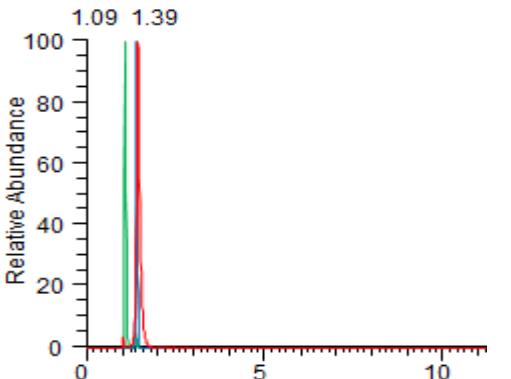
Isoleucine



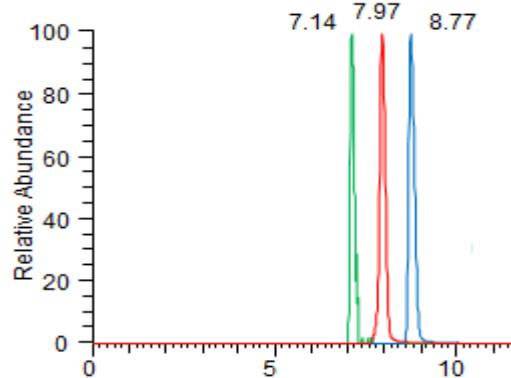
β Leucine



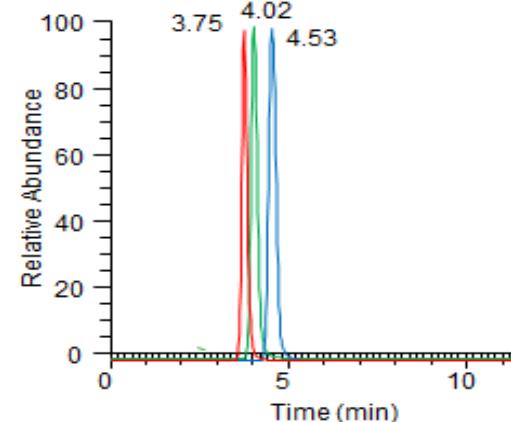
Norleucine



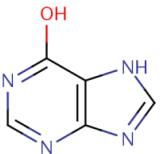
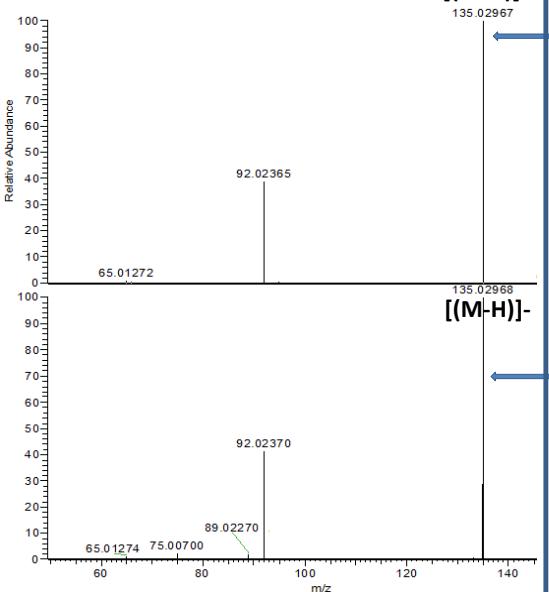
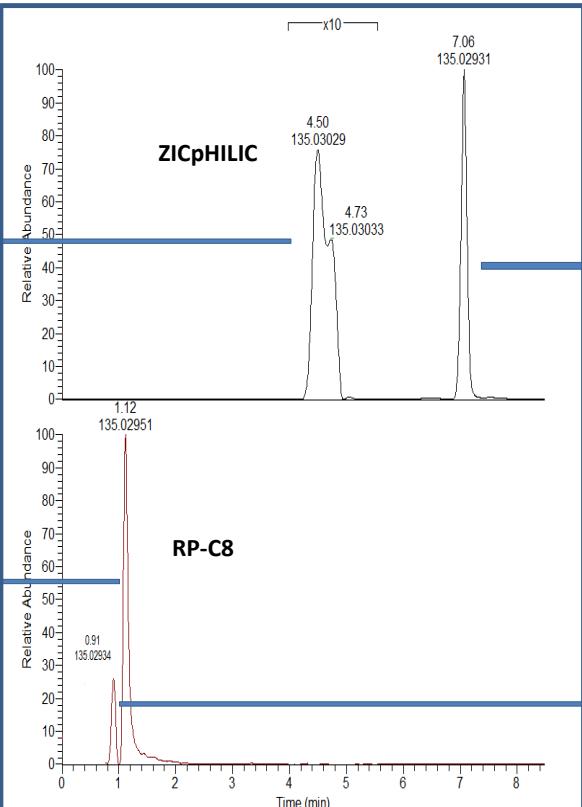
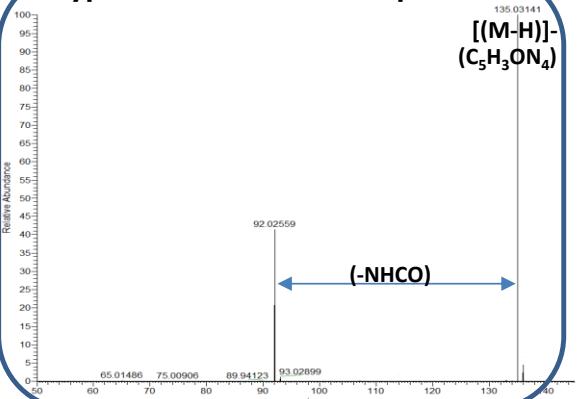
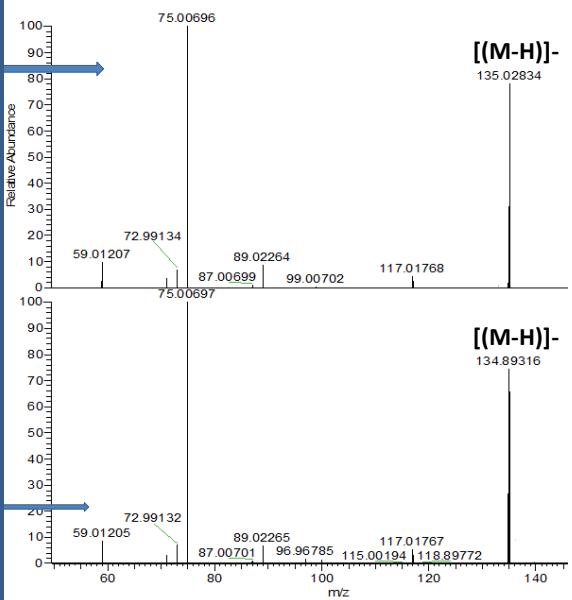
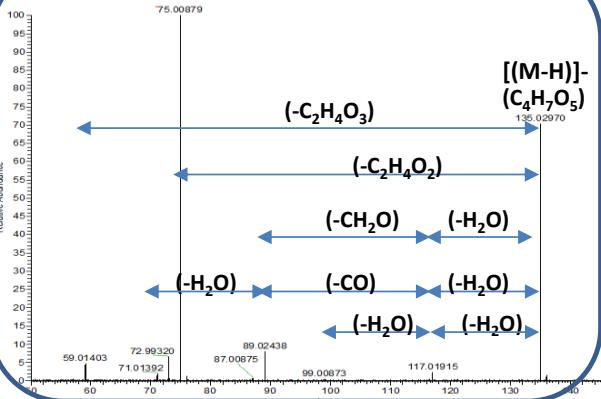
RP-C8



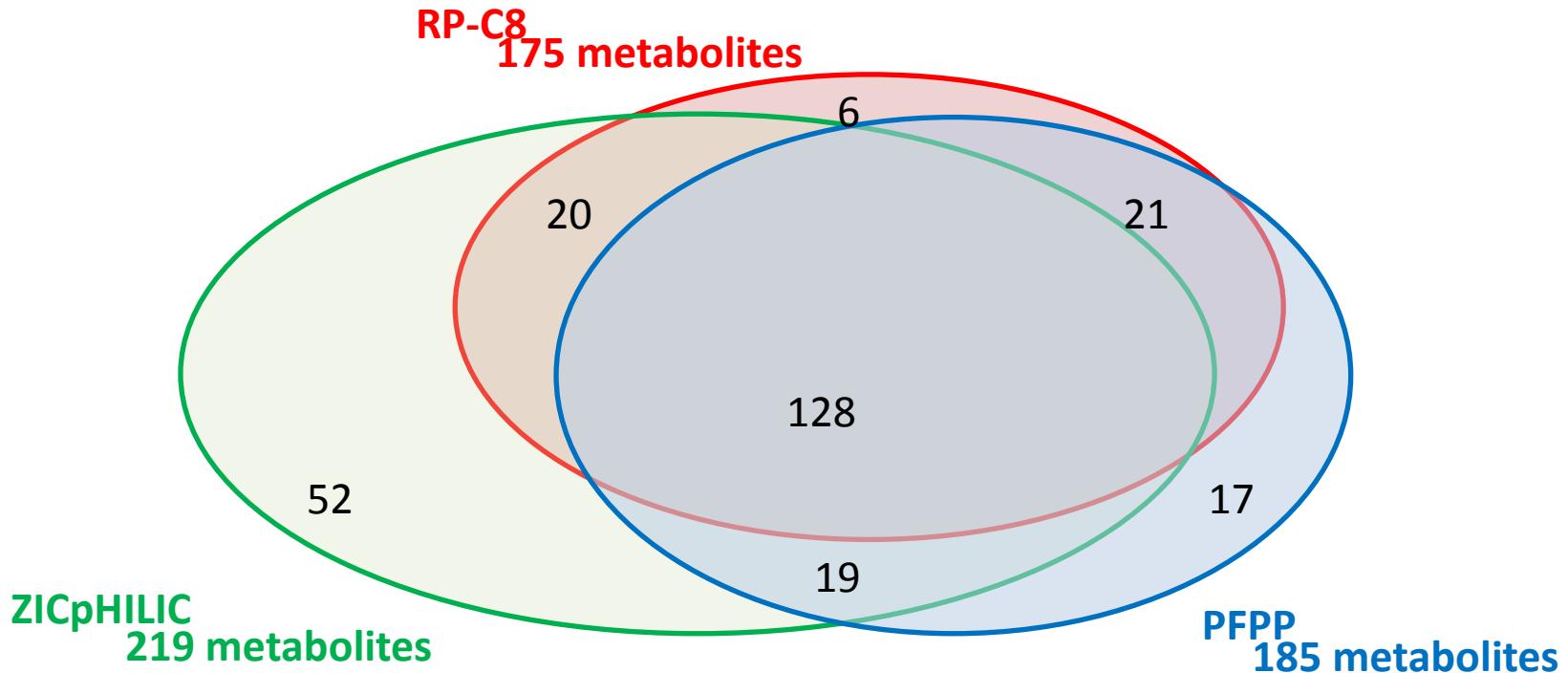
PFPP



ZICpHILIC

**LC/MS/MS analyses****LC/MS analyses of human plasma****Hypoxanthine reference spectra****LC/MS/MS analyses****Threonic acid reference spectra**

Annotation of the human serum metabolome using LC/HRMS



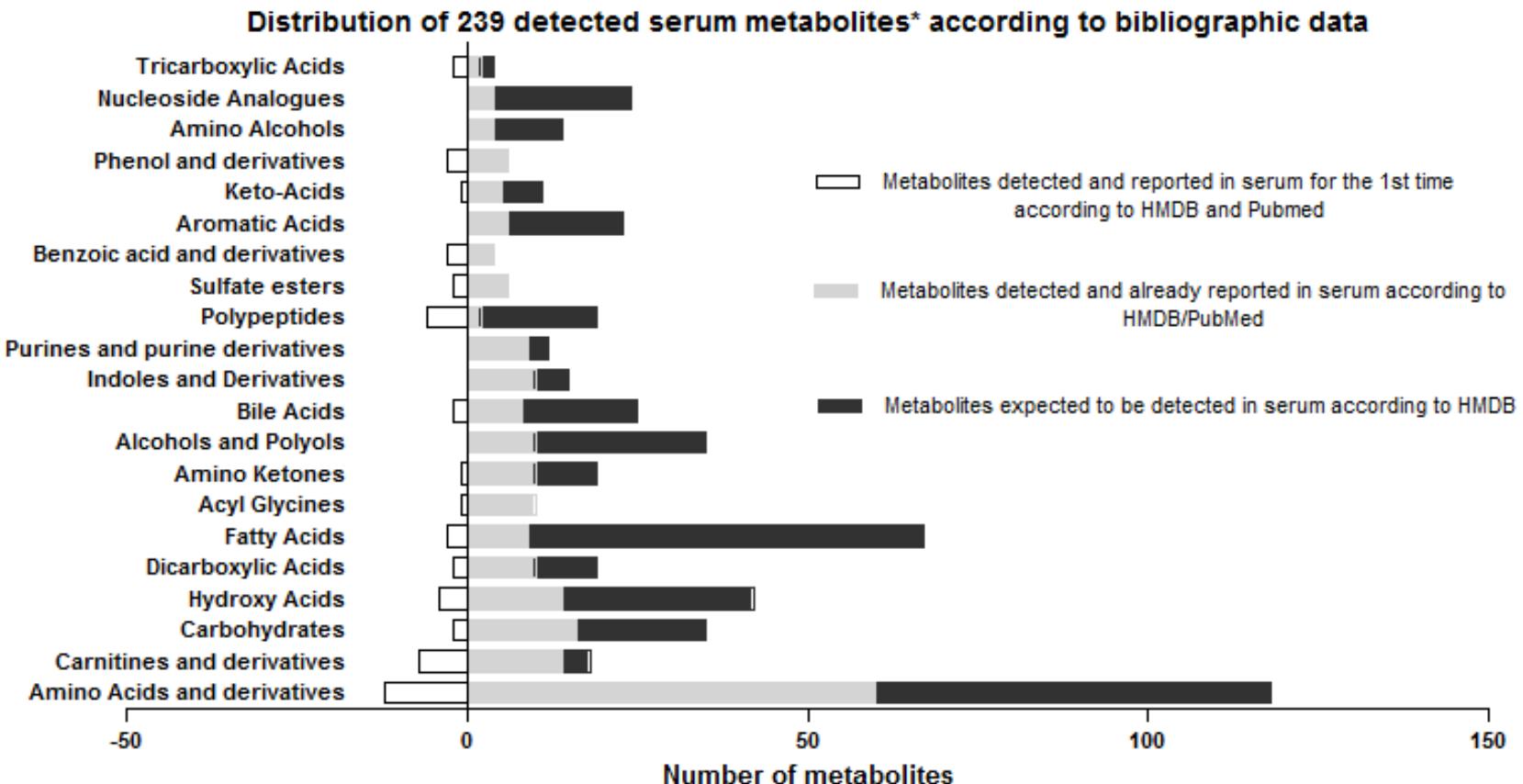
- 266 metabolites were distributed over 209 distinct accurate masses

- ESI(-): HILIC extended metabolome coverage 50% and the number of retained metabolites ($k>1$) 75%.

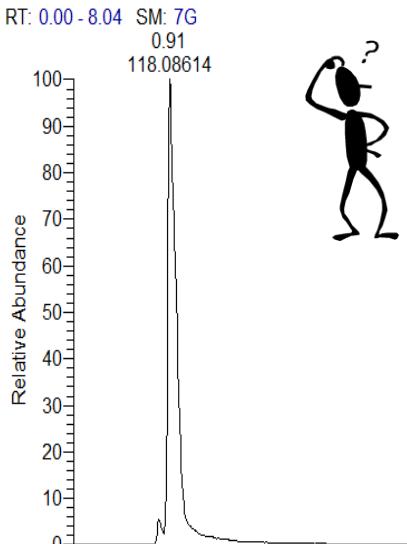
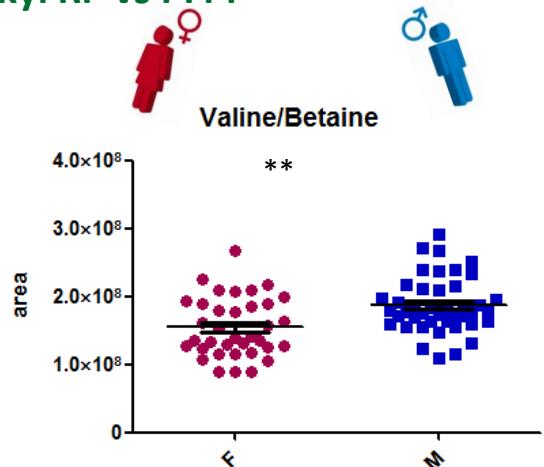
- ESI(+): PFPP conditions improved the retention up to almost 200% of metabolites detected.

- using HILIC (ESI-) and PFPP (ESI+) ensure the detection of 243 out of 266 metabolites in serum samples.

Annotation of the human serum metabolome using LC/HRMS

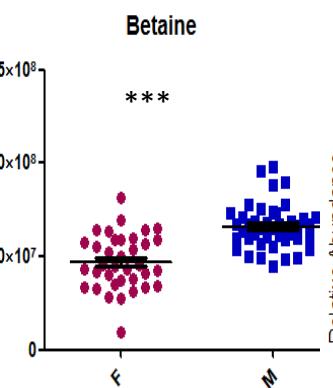
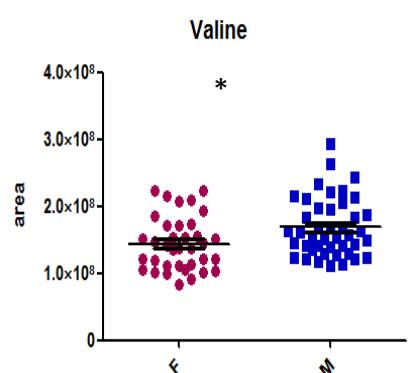


Alkyl RP vs PFPP



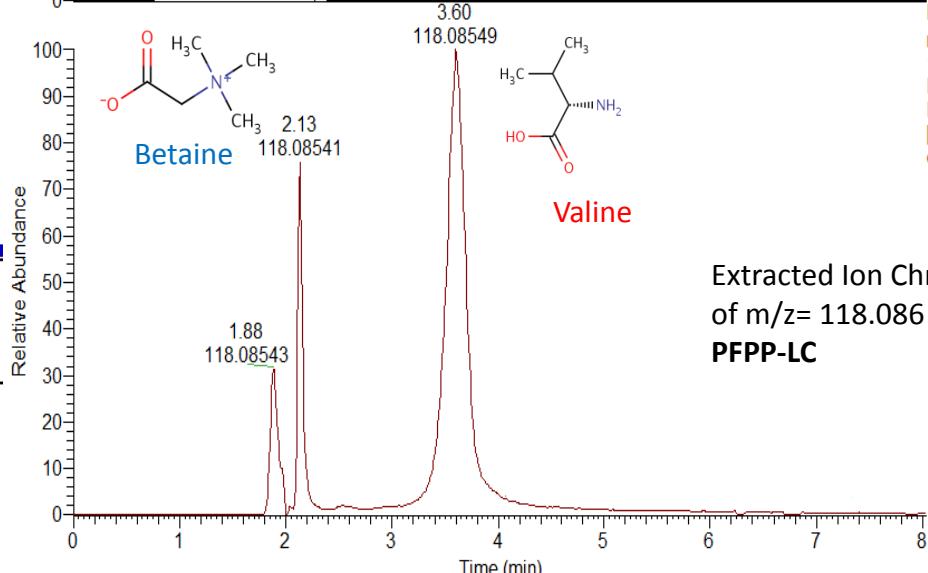
Extracted Ion Chromatogram
of $m/z = 118.086$ obtained in
RP-LC

NL: 3.36E7
 $m/z =$
118.08522-118.08710
F: FTMS {1;1} + p ESI
Full ms
[75.00-1000.00] MS
QC-Dil1_Pos_7



Proteinogenic
amino acids

Detoxification
reaction : liver,
kidney



Extracted Ion Chromatogram
of $m/z = 118.086$ obtained in
PFPP-LC

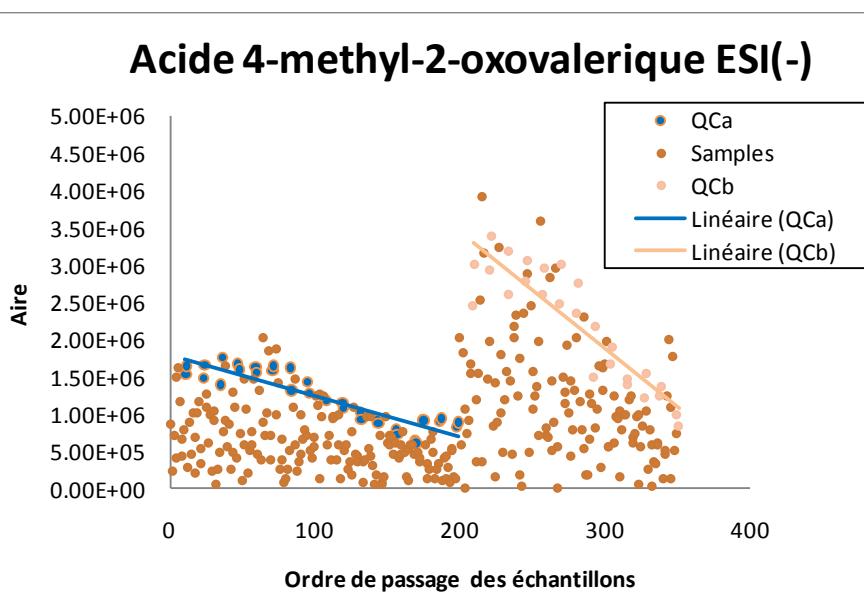
Toward databases of metabolic profiles: it is required to control analytical biais through design of experiment

- Randomization is required
- Batches of ~ 100 injections
- Blanck and QC samples

Some «reference» protocols are available:

Human plasma: Dunn WB, Nat. Protocols, 2011

Human urines: Want E., Nat. Protocols, 2010



Pool d'échantillons bio
préparés à 3 dilutions
et injectés en $n=3$ en
vue du traitement des
données

10 échantillons bio

QC-pool + EI_dil-8
QC-pool + EI_dil-8
QC-pool + EI_dil-8

QC-pool + EI_dil-4
QC-pool + EI_dil-4
QC-pool + EI_dil-4

QC-pool + EI_dil-2
QC-pool + EI_dil-2
QC-pool + EI_dil-2

blanc
QC-pool
QC-pool + EI

Ech bio
Ech bio
Ech bio
...

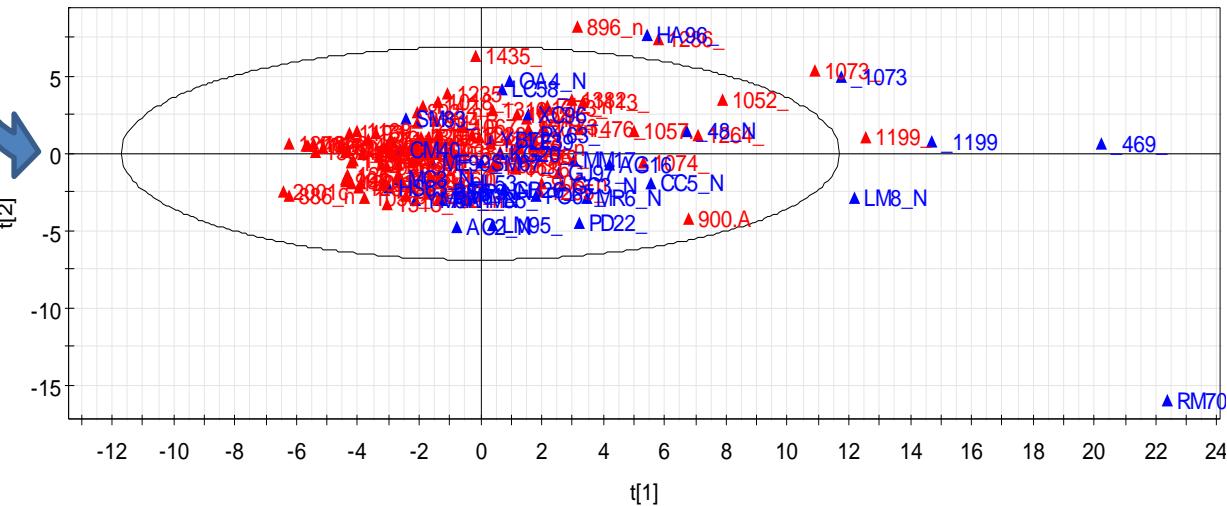
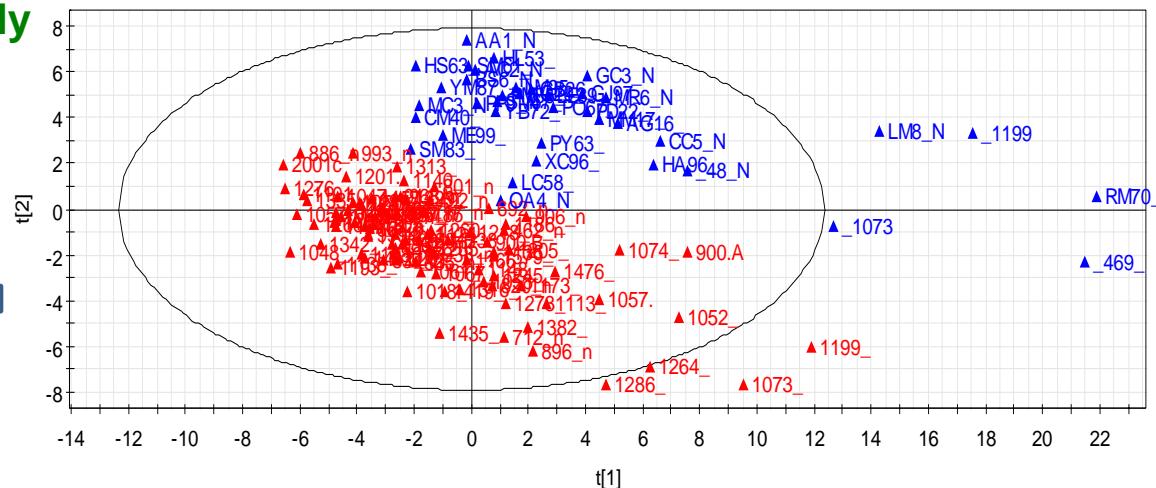
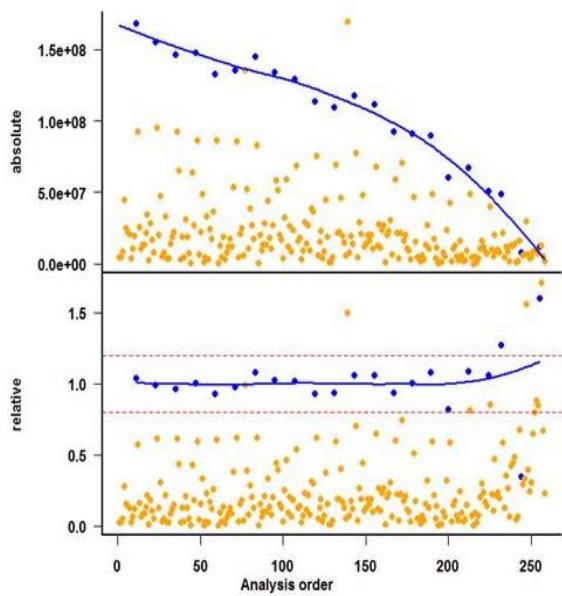
Ech bio
Ech bio

blanc
QC-pool
QC-pool + EI

Toward databases of metabolic profiles: it is required to control analytical biais through design of experiment

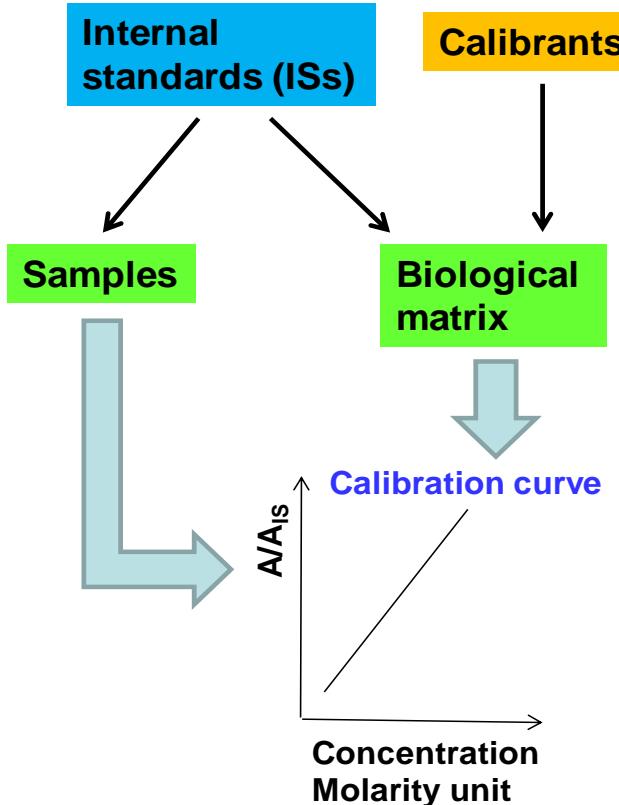
LOESS: Low Order non linear locally Estimated Smoothing Function

Cleveland, J. Am. Stat. Assoc. 1979
Dunn WB, Nat. Protoc. 2011



Quantitative metabolite profiling for large scale studies

Absolute quantification of metabolites



PLOS GENETICS

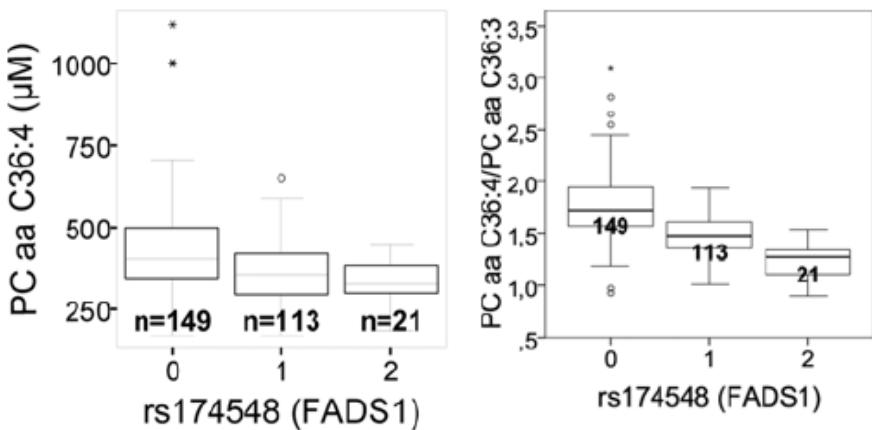
Genetics Meets Metabolomics: A Genome-Wide Association Study of Metabolite Profiles in Human Serum

Christian Gieger^{1,2}, Ludwig Geistlinger¹, Elisabeth Altmaier^{3,4}, Martin Hrabé de Angelis^{5,6}, Florian Kronenberg⁷, Thomas Meitinger^{8,9}, Hans-Werner Mewes^{3,10}, H.-Erich Wichmann^{1,2}, Klaus M. Weinberger¹¹, Jerzy Adamski^{5,6}, Thomas Illig¹, Karsten Suhre^{3,4*}

Quantitative measurement of 363 metabolites in 284 serum samples

«Genetically determined metabotypes»

Polymorphism in the *FADS1* (fatty acid delta 5 desaturase) gene



Metabolomics (2012) 8:757–760
DOI 10.1007/s11306-012-0462-0

MetaboLights: towards a new COSMOS of metabolomics data management

Christoph Steinbeck · Pablo Conesa · Kenneth Haug · Tejasvi Mahendraker ·
Mark Williams · Eamonn Maguire · Philippe Rocca-Serra · Susanna-Assunta Sansone ·
Reza M. Salek · Julian L. Griffin

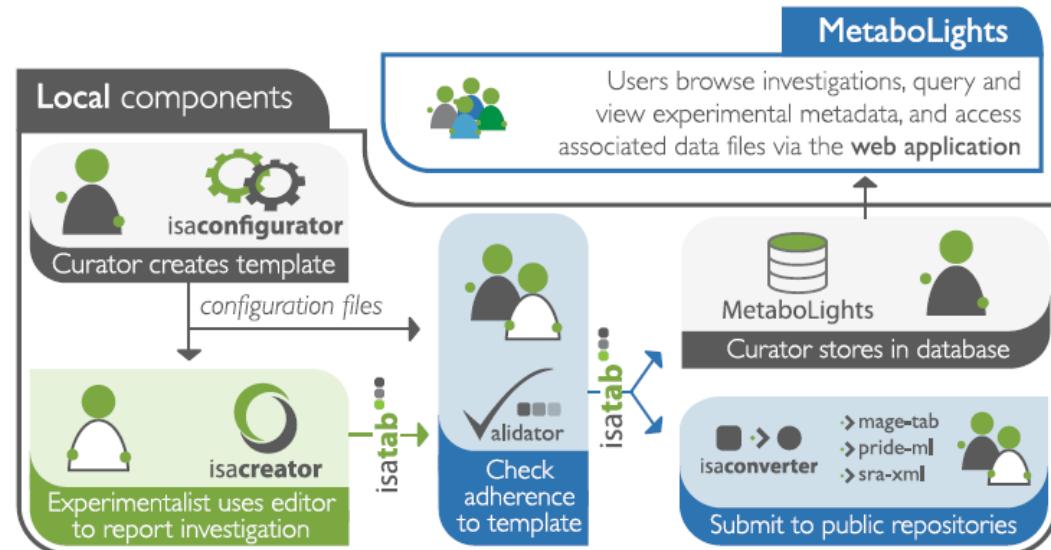
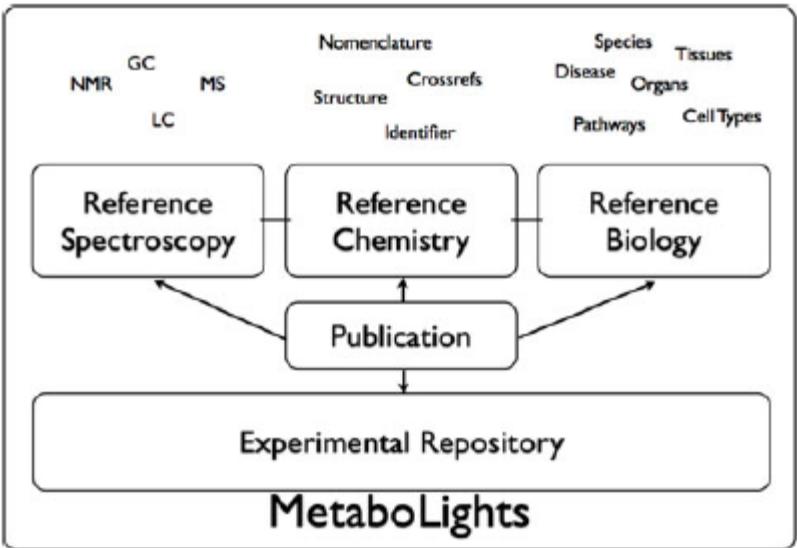


Fig. 2 MetaboLights data submission workflow

Conclusion

Apport significatif de la FT-MS dans la détection des métabolites (séparation des ions isobares) et dans leur identification (annotation/élucidation structurale).

Disponibilité de différents types de bases de données (bases spectrales de HRMS, de MS/MS, bases de données biochimiques et métabolomique). Les outils de normalisation et de quantification rendent possible la construction de bases de données de profils métaboliques.

Un des principaux enjeux: le partage des données concernant les composés inconnus. Pour cela, nécessité de standardisation des spectres MS/MS.

Très haute résolution (Orbitrap, FTICR) versus efficacité des acquisitions données dépendantes avec Q-TOF??

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